**RESEARCH ARTICLE** 

# Investigation of the landscape of RSV M protein with quercetin and acetylated quercetin in the search for antivirals

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### **PUBLISHED**

31 January 2026

#### **CITATION**

Seixas, L., Pedro, RP., et al., 2026. Investigation of the landscape of RSV M protein with quercetin and acetylated quercetin in the search for antivirals. Medical Research Archives, [online] 14(1).

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DOI

### ISSN

2375-1924

### **ABSTRACT**

Human Respiratory Syncytial Virus (hRSV) is a major causative agent of severe respiratory infections in infants, the elderly and immunocompromised individuals. The hRSV genome synthesizes eleven proteins, two of which are non-structural. Among its structural proteins, the matrix (M) protein plays a pivotal role in viral assembly and budding, making it a strategic target for the development of novel antiviral approaches. In this study, we investigated the interaction of the M protein with quercetin, a natural flavonoid, as well as with two acetylated derivatives, peracetylated auercetin and tetraacetylated auercetin. spectroscopic techniques in combination with computational biophysics methods. The results showed specific interactions in a 1:1 stoichiometry, characterized through moderate affinity and a predominance of hydrophobic forces. Acetylation enhanced conformational stability and increased the binding affinity of the complexes, with peracetylated quercetin exhibiting the strongest capacity for molecular recognition. Molecular dynamics simulations supported these findings, showing that chemical modification introduced additional contact points, including new hydrogen bonds, thereby conferring greater structural stability to the complexes. Overall, these findings contribute to the understanding of the molecular mechanisms that regulate the interaction between the M protein and small ligands and point to acetylated flavonoids as potential antiviral candidates capable of interfering with hRSV assembly and maturation

Keywords: Matrix protein; hRSV; quercetin; acetylated quercetin

### 1. Introduction

Human Respiratory Syncytial Virus (hRSV) is a major cause of lower respiratory tract infections (LRTIs) in children and elderly <sup>1</sup>. In pediatric populations, hRSV infection frequently manifests as bronchiolitis and is strongly associated with an increased risk of asthma later in life <sup>2</sup>. Among children under five years of age, hRSV remains one of the principal viral pathogens linked to pneumonia <sup>3</sup>. On a global scale, hRSV-associated acute lower respiratory infections (ALRIs) were estimated, in 2019, to account for around 33 million cases, leading to 3.6 million hospital admissions and 130,000 deaths <sup>4</sup>.

The hRSV belongs to the family Pneumoviridae and species *Orthopneumovirus hominis* <sup>5</sup>. It is an enveloped virus <sup>6</sup> with helical symmetry <sup>7</sup> and can display spherical, filamentous or pleomorphic forms <sup>8</sup>. Its genome is a nonsegmented, single-stranded, negative-sense RNA of approximately 15,200 nucleotides, encoding two nonstructural proteins (NS1 and NS2) and nine structural proteins. These include the surface glycoproteins F, G and SH, the nucleocapsid proteins N, P and L, the nucleocapsid-associated proteins M2-1 and M2-2 and the matrix protein M <sup>6</sup>.

The M protein consists of 256 amino acids arranged into two domains, N-terminal and C-terminal. The N-terminal domain adopts a horseshoe-like shape, while the Cterminal domain forms a flattened barrel structure. Both are largely composed of  $\beta$ -sheets and are linked with a flexible loop of 13 residues 9. The native form of M is a dimer, stabilized through the interaction of the N-terminal region of one protomer with the C-terminal region of the other. Hydrophobic regions at the interdomain connection point to a requirement for oligomerization and interactions with other proteins in order to achieve stability 10. Furthermore, phosphorylation of residue Thr205 modulates M oligomerization and regulates its nucleocytoplasmic trafficking 11. Functionally, M is a phosphorylated, non-glycosylated protein located just beneath the viral envelope, where it bridges the lipid bilayer to the nucleocapsid 12. During early stages of infection, M is directed to the host nucleus through Importin  $\beta$ 1, where it disrupts host transcription <sup>13,14</sup>. At later stages, it is exported back to the cytoplasm via CRM-1, where it associates with M2-1 and the ribonucleocapsid (RNP) and participates in the formation of viral inclusion bodies (IBs)  $^{15-17}$ . M interacts with glycoprotein G in the Golgi and subsequently with F at lipid rafts in the plasma membrane, coordinating the transition from viral transcription to assembly and budding  $^{18-21}$ .

Despite advances in vaccine development, treatment options for hRSV remain limited. At the moment, treatment consists of the use of monoclonal antibodies, such as palivizumab and nirsevimab and the broad-spectrum antiviral Ribavirin, which has a high cost, variable efficacy and various adverse effects <sup>22</sup>. Recently, three vaccines have been approved, two protein subunit vaccines, RSVPreF3 (Arexvy®, GSK) and RSVPreF (Abrysvo®, Pfizer) and one mRNA vaccine, mRNA-1345 (mRESVIA, Moderna) <sup>23</sup>. These vaccines are licensed for adults over 60 years of age, with Abrysvo® also approved for maternal immunization to provide passive protection to infants during their first six months of life <sup>24</sup>. However, the lack of broadly effective antivirals highlights the urgent need for new therapeutic strategies <sup>25</sup>.

In this context, natural compounds such as flavonoids have attracted attention as potential antiviral agents 26. Among them, quercetin (Q) is one of the most widely studied and has shown inhibitory effects against viruses rhinovirus influenza, including and immunodeficiency virus <sup>27–30</sup>. Quercetin has also demonstrated activity against RSV, reducing both infection and replication 31,32. Acetylated derivatives of quercetin have shown even greater potential, for example, quercetin pentaacetate has been reported to inhibit RSV adhesion in vitro more effectively than the unmodified compound 33. The acetylation is known to enhance cellular uptake, protect molecules from metabolic degradation and improve both bioactivity and bioavailability 34-40.

Based on this premise, we aimed to explore the interaction between the hRSV M protein and quercetin along with two of its acetylated derivatives, peracetylated quercetin (Q1) and tetraacetylated quercetin (Q2). The chemical structures of these compounds are illustrated in Figure 1. Comparing these derivatives, our objective is to understand how acetylation boosts the antiviral properties of quercetin and to aid in the strategic development of new hRSV therapeutic options.

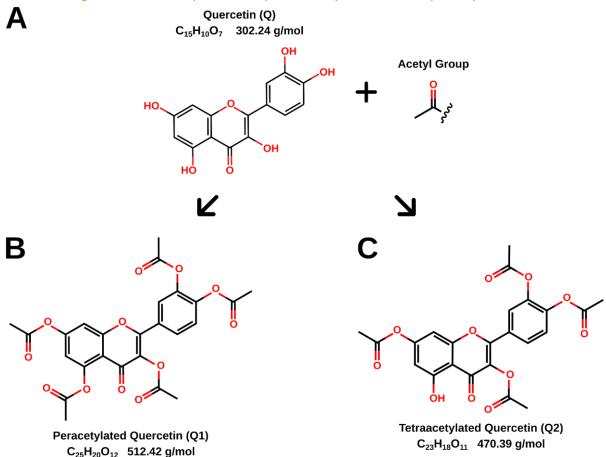


Figure 1: Chemical structures of quercetin and its acetylated derivatives. (A) Quercetin (B) Peracetylated quercetin (Q1), and (C) Tetraacetylated quercetin (Q2).

### 2. Materials and Methods

# a. EXPRESSION AND PURIFICATION OF RECOMBINANT M PROTEIN

The gene encoding the M domain of the M protein, which consists of 256 amino acid residues, was synthesized by DNA 2.0 (ATUM) and then inserted into the pD441-NHT-M expression vector. Plasmid transformation was performed in Escherichia coli BL21, using kanamycin (25  $\mu g/mL$ ) and chloramphenicol (37  $\mu g/mL$ ). Transformed colonies were inoculated into liquid LB broth supplemented with antibiotics and cultivated at 37 °C under agitation (100 rpm) for 4 hours. Recombinant expression was induced adding IPTG (0.4 mM) when the optical density at 600 nm (OD<sub>600</sub>) reached 0.7. The culture was subsequently incubated at 28 °C with agitation (120 rpm) for 16 hours. Following induction, cells were harvested via centrifugation (4500 rpm, 4 °C, 30 min) and the cell pellet was resuspended in lysis buffer (20 mM Tris-HCl, 500 mM NaCl, 5% glycerol, 0.1% Triton X-100, 1 mM  $\beta$ -mercaptoethanol, pH 7.5), supplemented with lysozyme (1 mg/mL) phenylmethylsulfonyl fluoride (PMSF, 0.5 mM). Cell lysis was carried out through sonication using six cycles of 30 s pulses (2 s on, 1 s off), interspersed with 40 s intervals in an ice bath. The lysate was subjected to centrifugation (17000 rpm, 4 °C, 40 min) and the resulting supernatant was filtered through 0.45 µm membranes prior to purification. Recombinant protein was initially purified through affinity chromatography using a Ni-NTA column pre-equilibrated with binding buffer (20 mM Tris-HCl, mM NaCl, 5 mM imidazole, 1 mM βmercaptoethanol, pH 8). Elution was performed using an imidazole gradient (10 mM to 1 M). To facilitate buffer exchange and remove contaminants, the eluate was subjected to size-exclusion chromatography. Protein purity was assessed using sodium dodecyl sulphate-polyacrylamide gel electrophoresis (SDS-PAGE, 15%) under denaturing conditions. Protein concentration was determined spectrophotometrically applying the Beer-Lambert Law, measuring absorbance at 280 nm and using a molar extinction coefficient of 27,390 M<sup>-1</sup> cm<sup>-1</sup> 9

### b. FLAVONOIDS

Quercetin (2-(3,4-dihydroxyphenyl)-3,5,7-trihydroxychromen-4-one) was purchased from Sigma-Aldrich (St. Louis, MO, USA). Quercetins acetylations were performed as previous described. Flavonoid stock solutions were prepared in a mix comprising 5% Milli-Q water and 95% ethanol (EtOH). Concentrations were determined in UV-Vis spectroscopy measurements were performed using a Cary-3E spectrophotometer (Varian, Palo Alto, CA) with quartz cuvettes featuring a 1 cm optical path length, according to the Beer-Lambert law <sup>41</sup>, using the specific molar extinction coefficients: quercetin (Q),  $\xi_{376} = 21,880 \, \text{M}^{-1} \, \text{cm}^{-1}$  ( $\lambda = 376 \, \text{nm}$ ) <sup>42</sup>; peracetylated quercetin (Q1),  $\xi_{266} = 29,305 \, \text{M}^{-1} \, \text{cm}^{-1}$  ( $\lambda = 266 \, \text{nm}$ ); and tetraacetylated quercetin (Q2),  $\xi_{268} = 21,720 \, \text{M}^{-1} \, \text{cm}^{-1}$  ( $\lambda = 268 \, \text{nm}$ ) <sup>38</sup>.

### c. FLUORESCENCE SPECTROSCOPY

The interaction between hRSV M protein and the ligands was assessed using fluorescence spectroscopy, monitoring the quenching of intrinsic tryptophan fluorescence upon ligand addition. Experiments were conducted on an ISS PC1 spectrofluorometer (Champaign, IL, USA) equipped with a quartz cuvette (1 cm optical path length) and a

NESLab RTE-221 thermal bath (Thermo Electron Corporation, USA). The excitation and emission slit widths were adjusted to 1.0 mm. Excitation was performed at 295 nm to selectively excite the tryptophan residues of the protein and emission spectra were recorded in the 300-500 nm range with ten accumulations. Fluorescence quenching assays were carried out via titrating the ligands (until 20  $\mu$ M) into a constant protein concentration (5  $\mu$ M). Measurements were performed at 288, 298 and 308 K. The protein remained in its native state throughout the studied temperature range, as the melting temperature of hRSV M protein is approximately 317 K  $^{43}$ . To minimize internal filtering effects and obtain a more reliable estimate of the emitted fluorescence, we applied the following equation (Eq. 1):

$$F_{corr} = F_{obs} x 10^{(A_{\lambda ex} + A_{\lambda em})/2}$$
 (1)

In this expression,  $F_{corr}$  denotes the corrected fluorescence intensity, whereas  $F_{obs}$  represents the experimentally measured value. The terms  $A_{\lambda ex}$  and  $A_{\lambda em}$  correspond to the absorbance values recorded at the excitation and emission wavelengths, respectively <sup>44</sup>.

To distinguish the quenching mechanisms and quantify binding events, the fluorescence data were analyzed with well-established models. First, the Stern-Volmer relationship (Eq. 2) was applied to determine whether quenching followed a dynamic or static profile:

$$\frac{F_0}{F} = 1 + k_q \tau_0[L] = 1 + K_{SV}[L] \tag{2}$$

where  $F_0$  and F are the fluorescence intensities in the absence and presence of ligand, respectively,  $\tau_0$  is the lifetime of the fluorophore in the absence of quencher, [L] is the ligand concentration,  $k_q$  is the bimolecular quenching constant and  $K_{SV}$  the Stern-Volmer constant.

Binding affinity and stoichiometry were then obtained using fitting the data to a double-logarithmic model (Eq. 3), which provides both the number of ligands bound per protein molecule (n) and the binding constant  $(K_b)$ :

$$log\left(\frac{F_0-F}{F}\right)=n(K_b)-nlog\left(\frac{1}{[L_t]}-\frac{F_0-F}{F_0}[P_t]\right) \tag{3}$$

where  $[L_t]$  and  $[P_t]$  correspond to the total ligand and protein concentrations, respectively.

Finally, to explore the driving forces of the interaction, thermodynamic parameters were derived using the Van't Hoff equation (Eq. 4):

$$\Delta G = -RT \ln(K_b) = \Delta H - T\Delta S \tag{4}$$

where R is the gas constant (1.98 cal  $mol^{-1} K^{-1}$ ), T is the

absolute temperature and  $K_b$  the binding constant. The resulting values of  $\Delta G$ ,  $\Delta H$  and  $T\Delta S$  allowed us to infer whether complex formation was mainly driven by enthalpic contributions, such as hydrogen bonding, or by entropic effects, typically associated with hydrophobic interactions.

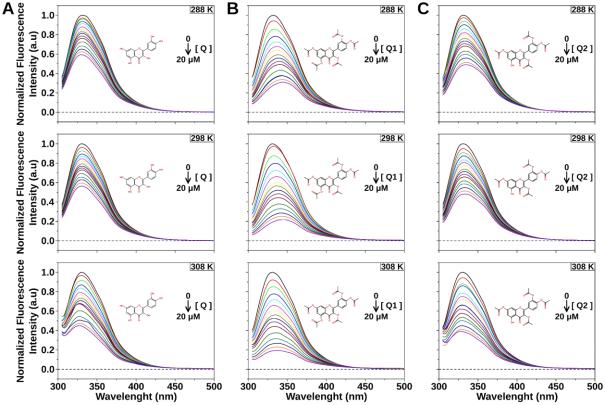
# d. MOLECULAR DOCKING AND MOLECULAR DYNAMICS

The interactions between the molecules and the M protein of the hRSV were assessed through molecular docking using the CB-Dock2 server 45,46, employing the crystallographic model of the M protein (PDB ID: 2VQP <sup>47</sup>) as the initial structure. Blind docking was performed for each compound to identify the most representative binding pose, characterized through the lowest interaction energy. The three-dimensional structures of the ligands and their corresponding topology files were generated using the CHARMM-GUI server 48,49. The highest-ranked docking model was employed as the initial structure for molecular dynamics simulations, conducted using the GROMACS 2020.3 package 50, with the CHARMM36 force field and the TIP3P water model <sup>51–53</sup>. Energy minimization was performed in two steps, 50,000 steps using the steepest descent algorithm, followed by 5,000 steps using the conjugate gradient method, both without positional restraints. Equilibration was carried out in two 100 ps phases, the first with restraints applied to both protein and ligand and the second without restraints. Production simulations were conducted for 450 ns with a 2 fs integration time step, at 298 K, 1 atm pressure and a physiological saline concentration of 0.15 M. Coordinates and energies were recorded every 100 ps. The Parrinello-Rahman barostat and Berendsen thermostat were employed 54,55. The simulation box was defined as a 10 Å buffer from the protein surface under periodic boundary conditions. Trajectory analysis was performed using GROMACSintegrated tools. The conformation exhibiting the lowest short-range interaction energy was selected as the representative structure of the M protein/ligand complex.

### 3. Results and Discussion

# $\alpha.$ LIGAND-INDUCED FLUORESCENCE QUENCHING OF THE HRSV M PROTEIN

To elucidate the interaction between the hRSV M protein and quercetin (Q) and its acetylated derivatives (Q1 and Q2), we performed intrinsic fluorescence experiments, exploiting selective tryptophan excitation at 295 nm (see Materials and Methods for details). The M protein exhibited a maximal emission around 340 nm, a value typical of tryptophan residues partially exposed to the solvent <sup>56,57</sup>. Upon addition of Q, a concentration-dependent decrease in fluorescence intensity was observed (Figure 2A). A similar pattern was noted for Q1 and Q2 (Figure 2B and C), indicating that all these compounds act as tryptophan fluorescence quenchers and, consequently, interact directly with the M protein.



**Figure 2:** Fluorescence emission spectra of the hRSV M protein in the presence of ligands (**A**) Q, (**B**) Q1, and (**C**) Q2. Measurements were performed with excitation at 295 nm, and each ligand was titrated stepwise from 0 to 20  $\mu$ M into a 5  $\mu$ M protein solution at 288 K, 298 K, and 308 K.

Although the emission maximum wavelength did not undergo significant shifts, the quenching efficiency varied among the compounds, reflecting differences in their interaction strengths. To further characterize the nature of this process, the data were analyzed using the Stern-Volmer equation (Eq. 2) across different temperatures (Figure 3).

The curves obtained for quercetin revealed that quenching increased with rising temperature, suggesting

a dynamic mechanism in which collisions between the fluorophore and quencher occur more frequently under warmer conditions. In contrast, the acetylated derivatives exhibited a distinct behavior, the quenching curves showed minimal temperature dependence, remaining nearly constant in some instances <sup>42</sup>. These findings point to a static quenching mechanism, associated with the formation of stable complexes with the M protein <sup>57</sup>.

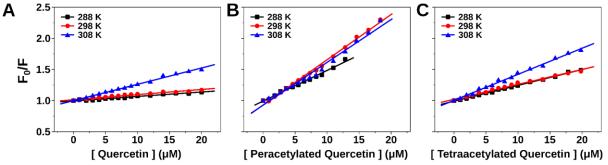
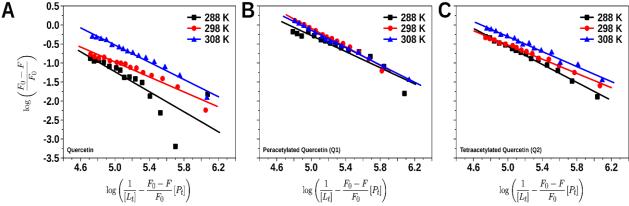


Figure 3: Stern-Volmer plots depicting the quenching of the intrinsic fluorescence of M protein as a function of ligand concentration: (A) Q, (B) Q1, and (C) Q2. The curves were recorded at 288, 298, and 308 K.

# b.BINDING CONSTANTS AND LIGAND SPECIFICITY OF THE HRSV M PROTEIN

The interaction of the hRSV M protein with quercetin and its acetylated derivatives was investigated using the double-logarithmic equation (Eq. 3)<sup>58</sup>, applied to the

fluorescence quenching data obtained experimentally (Figure 2). Fitting the experimental data with this model revealed consistent trends across the different ligands examined. These results are shown in Figure 4.



**Figure 4:** Double-logarithmic plots derived from fluorescence titrations of protein M with (A) Q, (B) Q1, and (C) Q2. Titrations were performed at different temperatures, 288 K (black squares), 298 K (red circles) and 308 K (blue triangles). The solid lines represent the linear fits, enabling the determination of the parameters n and  $K_b$ , as defined in Eq. 3.

The values of n and  $K_b$  are summarized in Table 1, enabling a systematic comparison of the distinct experimental conditions.

**Table 1:** Binding site number (n), binding constant  $(K_b)$  and Stern-Volmer constant  $(K_{SV})$  determined for quercetin and acetylated derivatives under different thermal conditions.

Ligands	T (K)	n	K <sub>b</sub> (х10 <sup>4</sup> ) М-1	Ksv (x103) M-1		
Quercetin (Q)	288	1.32 (± 0.27)	1.16 (± 0.40)	7.17 (± 0.36)		
	298	0.98 (± 0.06)	1.03 (± 0.11)	8.25 (± 0.45)		
	308	1.13 (± 0.04)	3.43 (± 0.22)	26.23 (± 0.56)		
Peracetylated quercetin (Q1)	288	1.06 (± 0.10)	5.50 (± 0.80)	49.33 (± 1.54)		
	298	1.22 (± 0.04)	8.50 (± 0.43)	72.75 (± 1.40)		
	308	1.14 (± 0.01)	7.55 (± 0.19)	68.81 (± 2.46)		
Tetraacetylated quercetin	288	1.20 (± 0.03)	3.33 (± 0.15)	25.39 (± 0.36)		
(Q2)	298	0.93 (± 0.02)	2.74 (± 0.12)	23.97 (± 0.70)		
	308	0.99 (± 0.03)	4.91 (± 0.26)	41.85 (± 0.88)		

Analysis of the binding constants (Kb) showed that all compounds displayed values on the order of 10<sup>4</sup> M<sup>-1</sup>. Such magnitudes are indicative of moderate binding affinity, which is nonetheless biologically relevant in the context of protein-ligand interactions 38,42. Despite sharing the same order of magnitude, the constants exhibited discernible differences that allowed the establishment of a relative affinity hierarchy: Q1 > Q2> Q <sup>38,59</sup>. Among the derivatives, peracetylated quercetin (Q1) consistently exhibited the highest binding constants across all tested temperatures, reaching 8.50 x 10<sup>4</sup> M<sup>-1</sup> at 298 K <sup>38</sup>. Tetraacetylated quercetin (Q2) showed intermediate values  $(2.74 - 4.91 \times 10^4 \text{ M}^{-1})$ , whereas unmodified quercetin displayed comparatively lower binding, particularly at lower temperatures, although a marked increase was observed at 308 K ( $K_b$ =  $3.43 \times 10^4 \,\mathrm{M}^{-1}$ ) <sup>38,42</sup>. This behavior is consistent with the dynamic nature of the interaction suggested by Stern-Volmer analyses 57,60.

These differences directly reflect the impact of chemical modifications on binding affinity. In the case of Q1, the enhanced binding constant suggests that acetylation favors recognition promotes M protein, most likely through strengthening hydrophobic contacts and decreasing the overall polarity of the molecule <sup>59,61</sup>. Although Q2 did not reach the affinity levels observed for Q1, its values remained consistently higher than those of unmodified quercetin, implying that partial acetylation also enhances the interaction. This reinforces the notion

that the degree of acetylation is a key determinant in modulating binding strength.

Another important aspect concerns the thermal behavior observed for quercetin, the sharp increase in both  $K_b$  and  $K_{SV}$  at 308 K indicates that binding is promoted by temperature-dependent mechanisms, compatible with dynamic quenching and transient contacts  $^{57,62}$ . In contrast, Q1 and Q2 displayed relatively stable Kb values across the studied temperature range, which is more consistent with static quenching processes involving the formation of stable protein-ligand complexes  $^{60,62}$ .

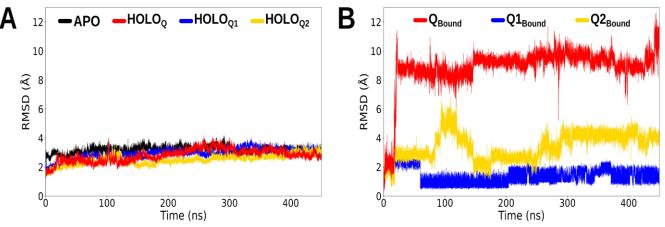
## c. COMPUTATIONAL EVALUATION OF INTERACTIONS

For the computational evaluation of ligand interactions, we selected the M protein protomer, which already provides the binding site to characterize the ligand interaction process.

In order to investigate the dynamic behavior and structural stability of the complexes formed between the M protein and the compounds Q, Q1 and Q2, molecular dynamics (MD) simulations were performed. These simulations were initiated from the lowest-energy conformations obtained through molecular docking analyses <sup>63,64</sup>. One of the key parameters used to monitor the stability of the systems over the course of the simulations was the root-mean-square deviation (RMSD), calculated based on the coordinates of the protein

backbone atoms  $^{63,65,66}$ . For the ligands, RMSD values were determined through considering all heavy atoms  $^{64,67}$ . The RMSD profiles of the unbound M protein (APO) and the protein in complex with the three ligands (HOLO $_{\rm Q}$ , HOLO $_{\rm Q1}$  and HOLO $_{\rm Q2}$ ) are presented in Figure 5A. The positional deviations of the ligands relative to their respective binding sites within the M protein are shown in Figure 5B. In general, the RMSD data indicates that the presence of the ligands does not compromise the conformational stability of the M protein throughout the

simulation timescale (Figure 5A) <sup>63</sup>. Regarding the ligands themselves, all three compounds maintained stable interactions with the protein for the majority of the simulated trajectory (Figure 5B). However, the acetylated derivatives (Q1 and Q2) exhibited more consistent stability when compared to the unmodified quercetin. These findings agree with experimental data obtained via fluorescence spectroscopy (Figure 2), which also suggest a more favorable interaction between the M protein and the acetylated derivatives <sup>38</sup>.



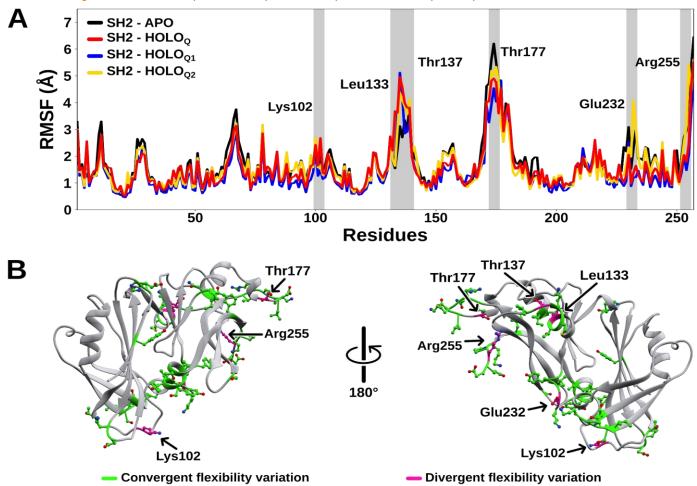
**Figure 5:** RMSD analysis of the hRSV M protein in different ligand-binding. **(A)** Temporal evolution of the RMSD for the APO system (ligand-free, black), HOLOQ (red), HOLOQ1 (blue), and HOLOQ2 (yellow). **(B)** RMSD of the ligands relative to their initial positions within the M protein binding site, used to assess the stability of the complexes.

As a complement to the structural stability analysis via RMSD, we performed root mean square fluctuation (RMSF) calculations to assess the local flexibility of the M protein residues throughout the simulations <sup>64,66</sup>. This approach enabled the identification of regions undergoing dynamic alterations in response to ligand binding. Mean RMSF values were obtained from three independent replicates for each system. Although Figure 6A presents the RMSF profiles directly, the comparative assessment between APO and HOLO states was still based on the residue-wise differences (APO - HOLO). In Figure 6A, these RMSF profiles are color-coded for clarity, red for HOLO<sub>Q</sub>, blue for HOLO<sub>Q1</sub> and yellow for HOLO<sub>Q2</sub>. Higher RMSF values indicate increased flexibility in a given state.

In summary, most residues exhibited reduced mobility in the presence of ligands, suggesting a stabilizing effect induced through complex formation <sup>64,66</sup>. A specific region spanning residues Leu133 to Thr137 displayed increased

flexibility across all HOLO conditions, which may reflect local conformational adjustments driven via ligand interactions <sup>10,68</sup>.

The highlighted residues (Figure 6B) show the most marked differences between the unbound and bound states. These residues were identified using the RMSF difference (APO - HOLO) and those showing consistent variations (RMSF differences > 0.5 Å) across all complexes are marked in green, while those exhibiting ligand-specific behaviors are shown in pink. Among these, Lys102, Leu133, Thr137, Thr177, Glu232 and Arg255 are notable for their differential responses to the ligands, potentially reflecting specific binding interactions or allosteric effects mediated by acetylation patterns in each derivative. Αll analyzed residues comprehensively listed in Table S1. A complete list of residues categorized according to these criteria is available in the Supporting Information (Table S1).

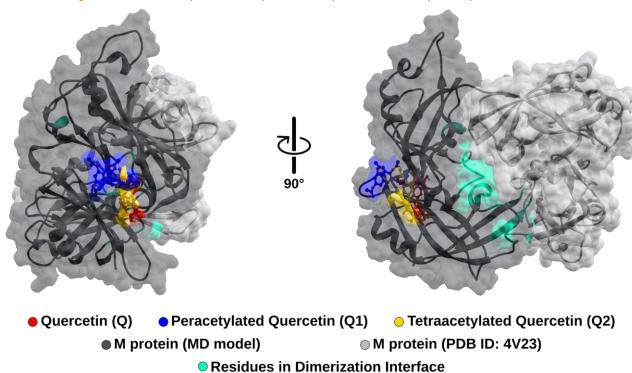


**Figure 6:** Flexibility of the M protein in complexes with quercetins. **(A)** RMSF profiles for APO state (black), complex with Q (HOLO $_Q$ , red), Q1 (HOLO $_Q$ 1, blue) and Q2 (HOLO $_Q$ 2, yellow). **(B)** Changes mapped onto the M protein, green residues represent convergent responses to ligand binding, and magenta residues denote divergent flexibility patterns.

То better understand the potential functional consequences of the flexibility variations observed in the molecular dynamics simulations, we performed a comparative analysis between the protomer model of the M protein used in this computational part and its crystallographic dimeric structure (PDB ID: 4V23 10). The purpose of this comparison was to determine whether the residues showing significant fluctuations in the RMSF profiles (Table S1) correspond to regions previously reported as part of the dimerization interface. According to previous studies, the main residues forming this interface include Lys60, Thr63-Leu68, Ile71, Lys93, Leu96-lle98, lle104, Ala105, Gly129, Leu130, Leu133, Tyr142, Ser144, Ile159, Gly161, Val163, Gly228, Gly229, Ala231, Glu232 and Arg255. Notably, several

of these residues also displayed pronounced differences in flexibility between the APO and HOLO forms, as indicated in Table S1  $^{10}$ .

Structural inspection of the molecular dynamic complexes revealed, however, that the ligands do not bind directly to the dimerization interface. As illustrated in Figure 7, the binding pocket is located in a topologically distinct region that remains structurally connected to the interface through adjacent secondary structure elements. This spatial arrangement suggests that ligand binding may modulate the overall dynamics of the M protein through an allosteric mechanism, propagating conformational changes toward the dimerization interface <sup>69,70</sup>.



**Figure 7:** Structural comparison between the protomer model of the M protein (dark gray surface) and the crystallographic dimer (light gray surface). The binding sites of Q (red), Q1 (blue), and Q2 (yellow). Residues forming the dimerization interface are highlighted in aquamarine.

## d. HYDROPHOBICITY AS A DETERMINING FACTOR IN MOLECULAR INTERACTION

Fluorescence experiments also enabled the characterization of the thermodynamic parameters governing the interaction between protein M and the different ligands. Based on the Van't Hoff equation (Eq.

4), the corresponding plots were constructed (Figure S2), from which the values of  $\Delta H$ ,  $\Delta G$  and  $T\Delta S$  were derived at distinct experimental temperatures <sup>71</sup>. A summary of these results is presented in Table 2.

**Table 2:** Thermodynamic parameters ( $\Delta H$ ,  $\Delta G$  and  $T\Delta S$ ) associated with the interaction of protein M with Q, Q1, and Q2 at 288, 298, and 308 K.

Ligands	T (K)	ΔH kcal Mol-1	ΔG kcal Mol <sup>-1</sup>	TΔS kcal Mol <sup>-1</sup>
Quercetin	288	9.40 (± 3.68)	-5.35 (± 1.84)	14.35 (± 10,72)
	298		-5.47 (± 0.58)	14.47 (± 7,31)
	308		-6.40 (± 0.40)	15.40 (± 7.12)
Peracetylated quercetin (Q1)	288	2.92 (± 0.48)	-6.25 (± 0.91)	9.16 (± 1.50)
	298		-6.72 (± 0.34)	9.64 (± 1.58)
	308		-6.87 (± 0.17)	9.80 (± 1.60)
Tetraacetylated quercetin	288	3.57 (± 0.43)	-5.95 (± 0.27)	9.53 (± 1.60)
(Q2)	298		-6.05 (± 0.26)	9.62 (± 1.60)
	308		-6.61 (± 0.35)	10.20 (± 1.78)

The negative  $\Delta G$  values confirm the spontaneous nature of the association process under all experimental conditions  $^{72}$ . However, the principal contribution to complex stabilization is entropic (T $\Delta S$ ), whose magnitude markedly outweighs the enthalpic term ( $\Delta H$ ). Such behavior is characteristic of processes predominantly driven by hydrophobic interactions, in which dehydration of the protein-ligand interface increases the overall entropy of the system  $^{73,74}$ .

When comparing the compounds, Q1 and Q2, display more consistent  $T\Delta S$  profiles across the temperatures investigated, whereas unmodified quercetin exhibits greater variation, consistent with the dynamic quenching mechanism proposed earlier. Graphical analysis (Figure 8) supports this distinction, indicating that acetylation

strengthens protein-ligand association not only by enhancing hydrophobic contributions but also introducing novel, specific contact points  $^{75,76}$ .

In Figure 8, interaction diagrams generated with LigPlot, derived from the most representative conformations of the molecular dynamic trajectories, reveal that quercetin binds primarily through hydrophobic contacts, whereas the acetylated derivatives additionally establish hydrogen bonds, Q1 with lle195 and Q2 with Ser117. These supplementary contacts account for the experimentally observed affinity hierarchy (Q1 > Q2 > quercetin) and are consistent with clustering analyses of the trajectories (Figure S1B), which revealed increased conformational stability for complexes containing the acetylated derivatives  $^{36,59,75}$ .

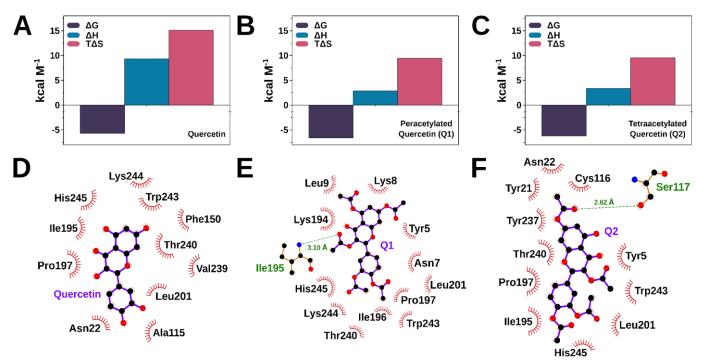


Figure 8: Bar plots of  $\Delta H$ ,  $\Delta G$  and  $\Delta S$  values for Q (A), Q1 (B), and Q2 (C) at 288, 298, and 308 K. Diagrams generated via LigPlot from the most representative conformations of the molecular dynamic simulations: Q1 (D), Q1 (E), and Q2 (F). Hydrophobic interactions (green), with additional hydrogen bonds observed for Q1 (Ile195) and Q2 (Ser117).

### 4. Discussion

The comparative results showed that Q1 was the most effective guencher regardless of temperature, subsequent to Q2, whereas the quercetin scaffold had a significantly lower quenching efficacy. According to this hierarchy, acetylation probably increases hydrophobic surface area and permits more nonpolar interactions with the M protein, which increases ligand-protein contact <sup>33,38</sup>. In essence, the distinct quenching regimes observed dynamic quenching predominating for quercetin and a substantial static component for the acetylated derivatives 77 indicate that these molecules engage the M protein through divergent interaction modes.

These physicochemical distinctions are relevant when contextualized within the multifunctional role of the M protein in the RSV life cycle. As established in structural and cell-based studies, M functions as the central organizer of virion assembly, bridging the RNP complex to the viral membrane and coordinating interactions with the envelope glycoproteins F and G 68. M also undergoes regulated nucleocytoplasmic trafficking, with nuclear accumulation suppressing host transcription early in infection and subsequent cytoplasmic export enabling virion morphogenesis 78,79. Because both processes rely on precisely tuned conformational transitions, the formation of more stable ligand-bound complexes, as observed for Q1 and Q2, may influence the conformational dynamics required for M to execute these regulatory functions.

The enhanced quenching efficiency and reduced temperature dependence of Q1 and Q2 directly reflect stronger and more thermodynamically stable interactions, consistent with the predominance of hydrophobic forces that are also central to M oligomerization and membrane association <sup>79,80</sup>. Thus, acetylated derivatives may

preferentially engage hydrophobic patches on M that participate in structural rearrangements essential for viral assembly, offering a mechanism rationale for the differential quenching behaviors detected <sup>81</sup>.

Given that M orchestrates multiple macromolecular interactions, including contacts with the nucleocapsid, glycoproteins, lipid rafts, and cytoskeletal components 79,80 even subtle alterations in its conformational plasticity may disrupt the efficiency of virion assembly and budding 68. The high stability of Q1 and Q2 bound complexes suggests that such interactions may persist under diverse intracellular conditions, with the potential to modulate M dimerization kinetics or the recruitment of assembly factors 62,68. The binding parameter n remained close to 1 under all experimental conditions, indicating that each M protein chain preferentially accommodates a single ligand molecule 60,82. This supports the existence of a specific and well-defined binding mechanism, rather than a diffuse interaction involving multiple low-affinity sites 60.

Our MD showing reduced flexibility in residues implicated in dimerization, despite the absence of direct ligand contact <sup>69,83</sup>. These findings are consistent with an allosteric restriction model, in which hydrophobic chemical modifications reinforce the transmission of dynamic constraints across the protein <sup>38,61</sup>. Because M dimerization is indispensable for filamentous virion formation <sup>79,80</sup>, perturbing this conformational cycle could hinder productive budding.

Beyond structural roles, recent work demonstrates that M also acts as an antagonist of the type I interferon response through interacting with the host adaptor RACK1 (Receptor for activated C kinase 1), an adaptor protein recruited via specific signaling molecules to

regulate diverse cellular processes and viral infection <sup>78</sup>. This immunomodulatory activity requires discrete conformational states and stable protein—protein interfaces. Ligands that restrict M flexibility even when bound allosterically could impair ability of M protein to engage RACK1, thereby attenuating its ability to suppress antiviral signaling pathways.

All our findings suggest that hydrophobicity is an essential attribute in the binding of the hRSV M protein to ligands. However, acetylation does more than enhance binding strength it qualitatively reshapes the interaction landscape through introducing additional anchoring points that stabilize the ligand-protein complex 75,76. These combined effects position acetylated quercetin derivatives as promising scaffolds for rational antiviral design, with the potential to simultaneously disrupt multiple structural and regulatory functions of M, including dimerization and membrane association.

### 5. Conclusion

Quercetin and its acetylated derivatives, Q1 and Q2, were found to interact with the M protein of the human respiratory syncytial virus with moderate yet biologically relevant affinities. These interactions are mainly driven by hydrophobic forces, and acetylation appears to play a critical role in enhancing conformational stability and strengthening intermolecular contacts. Among the analyzed compounds, Q1 exhibited the most stable complex formation and the highest number of molecular contacts, suggesting that the acetyl substitution pattern directly affects the molecular recognition of the M protein.

Molecular dynamics simulations showed that ligand binding stabilizes regions associated with the dimerization interface, suggesting a potential allosteric mechanism modulating M protein dimerization. This behavior aligns with previous experimental evidence demonstrating the virucidal activity of quercetin derivatives and their ability to downregulate F protein expression <sup>33</sup>.

Our data indicates that quercetin and its acetylated derivatives may act at multiple stages of the hRSV replication cycle. Further antiviral studies focusing on the effects of these compounds on M protein-mediated viral assembly and particle release will be essential to validate the functional relevance of these interactions and to guide the rational design of optimized flavonoid-based therapeutics against hRSV.

### **Acknowledgement**

We would like to express our sincere gratitude to the anonymous reviewers for their careful suggestions and constructive criticism, which undoubtedly contributed to the enhancement of this work. This study was supported by the National Council for Scientific and Technological Development (CNPq: 310898/2021-8, 317157/2023-0 and 305480/2024-3), the São Paulo Research Foundation (FAPESP: 2019/04646-9, 2022/01492-3 and 2022/13050-5) and the Research Pro-Rectory of São Paulo State University (PROPe/UNESP: 5545). This study made use of NMRbox: National Center for Biomolecular NMR Data Processing and Analysis, a Biomedical Technology Research Resource (BTRR), which is supported by NIH grant P41GM111135 (NIGMS).

### **Declaration of Competing Interest**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper as potential competing interests.

### **Author Contributions**

L.S., J.M.S., J.S.B. and M.S.S. performed experimental studies. R.P.P. and J.V.P. performed computer studies. L.O.R., M.A.F., Í.P.C. and F.P.S. designed the project and supervised the experimental and computer work. L.S., R.P.P., J.M.S., J.S.B., L.O.R., M.A.F., Í.P.C. and F.P.S. wrote the manuscript. L.S. and R.P.P. contributed equally as cofirst authors. All authors contributed to the analysis and discussion and provided critical feedback.

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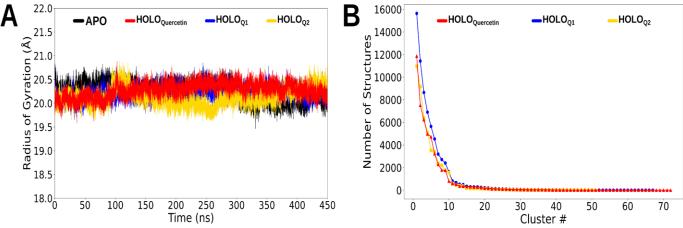
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### **Supporting Information**

**Table S1:** RMSF differences when comparing protein M (APO) with complexes (HOLO) Q, Q1, and Q2. Positive  $\Delta$ RMSF (> 0.5 Å) indicate increased rigidity in the interaction, and negative (< -0.5 Å) suggest flexible region. Green residues denote structurally sensitive regions potentially critical for the stability of protein. Not highlighted residues reflect ligand-specific interactions

specific interactions	HOLO <sub>Quercetin</sub>			HOLO <sub>Q1</sub>			HOLO <sub>Q2</sub>		
	Tyr5	Lys102	Leu133	Tyr5	Thr134	Met135	Tyr5	Asn22	Thr134
$\Delta$ RMSF < -0.5 Å	Thr134	Met135	Lys136	Lys136	Thr 1 <i>77</i>		Met135	Lys136	Thr 137
	Thr137	Leu138	Thr 1 <i>77</i>				Leu138	Glu232	Arg255
	His10	Lys26	Asp27	His10	Glu11	Lys26	His10	Glu11	Ser64
	Asp28	lle63	Ser64	Asp27	Lys51	Lys61	Thr65	Pro66	Lys67
	Thr65	Lys67	Ser70	Gln62	lle63	Ser64	Ser169	Val170	Arg171
	Arg72	Thr108	Ser156	Thr65	Pro66	Lys67	Asn172	Asp174	Thr 1 <i>77</i>
	Val170	Arg171	Asn172	Gly68	Pro69	Ser70	Glu179	Asn180	Thr182
	Lys173	Asp174	Leu175	Cys92	Ala93	Asn94	Thr183	Phe186	lle190
	Asn176	lle190	Thr191	Val95	Asp98	Lys102	Thr191	Asn192	Asn210
	Asn192	Gly228	Ala229	Tyr105	Asp106	Thr108	Gly228	Tyr230	Phe250
	Tyr230	Leu231	Glu232	Thr137	lle153	Val154	Glu256	Asp257	
	Lys233	Glu234	Ser235	Thr155	Ser156	Ser169			
$\Delta$ RMSF > 0.5 Å	lle236	Tyr237	Tyr238	Val170	Arg171	Asn172			
	Val239	Thr241	Phe250	Lys173	Asp174	Leu175			
	Lys253	Pro254	Arg255	lle190	Thr191	Asn192			
	Glu256	Asp257		Gly228	Ala229	Tyr230			
				Leu231	Glu232	Lys233			
				Ser235	lle236	Tyr237			
				Tyr238	Val239	Thr240			
				Thr241	Asn242	Lys244			
				Phe250	lle252	Lys253			
				Pro254	Arg255	Glu256			
				Asp257					



**Figure \$1:** Structural comparison of the M protein in its unbound and ligand-bound forms. **(A)** Radius of Gyration of the M protein in its APO (black) form and in HOLO states Q (red), Q1 (blue), and Q2 (yellow). **(B)** Distribution of the predominant conformational states of the M protein/ligand complexes, obtained via clustering analysis of the simulation trajectories.

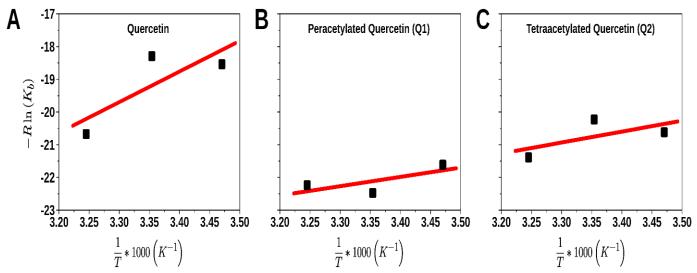


Figure S2: Van't Hoff plots derived from fluorescence measurements of the hRSV M protein in the presence of three ligands: Q (A), Q1 (B), and Q2 (C). Red lines represent the linear fitting of the experimental data, shown as black dots.