



RESEARCH ARTICLE

Iron and Micronutrient Deficiencies as Drivers of Carcinogenesis and Therapeutic Resistance in Cancer: An In-Depth Clinical and Physiological Three-Part Review

Part I—Iron Deficiency, Insufficiency, Excess, and Neoplasia

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ABSTRACT

The 2019 Nobel Prize in Physiology or Medicine underscored a fundamental insight: cellular oxygen-sensing and hypoxia pathways orchestrate key biological responses, including the initiation and progression of cancer. Building on this framework, this review translates the core science of oxidative stress and hypoxia-driven molecular alterations into practical clinical strategies. Latent iron deficiency and iron overload are both common, frequently underdiagnosed, and potent mediators of redox activity capable of driving carcinogenesis.

This manuscript presents a mechanistic narrative review of how iron deficiency and iron overload generate reactive oxygen species (ROS), promote tumor initiation, progression, and resistance to therapy.

Emphasis is placed on equipping clinicians who manage patients with hereditary cancer predisposition, subclinical premalignant lesions, or early biomarkers of oncogenesis to recognize iron imbalance earlier and implement targeted surveillance and timely intervention before developing neoplastic lesions.

Introduction

Part I of this three-manuscript series examines the fundamental pathophysiology of oxidative stress and its role in oncogenesis. In this installment, we focus on how imbalances in essential micronutrients exacerbate oxidative stress and how cellular hypoxia-driven signaling pathways both initiate and sustain pro-carcinogenic processes, often progressing silently below the threshold of routine clinical detection. Attention to this issue was first drawn in 1992 by Dr Victor Herbert, who asserted, "Everyone Should be Tested for Iron Disorders."

¹ In 2018, insights from Dr Esa Soppi ² of Finland reinforced the enduring importance of early detection and management of subclinical iron deficiency, a potent source of oxidative stress, as well as deficiencies of other micronutrients and their contribution to prolonged, clinically undetected neoplasia.

Aim and scope of Part I

In this first installment, we focus on how iron insufficiency and iron overload perturb redox balance and oxygen-sensing pathways, particularly reactive oxygen species (ROS), hypoxia-inducible factors (HIF-1 and HIF-2), nuclear factor κ B (NF- κ B), and nuclear factor erythroid 2-related factor 2 (Nrf2), to promote tumor initiation, progression, and resistance to therapy. This is a mechanistic narrative review centered on adult human malignancies in which iron imbalance is plausibly involved in carcinogenesis or treatment response; it synthesizes experimental, translational, and selected clinical data rather than providing a systematic trial meta-analysis. Other micronutrient deficiencies (folate, vitamin B12, vitamin B6, and vitamin D) and their epigenetic consequences are introduced only when they directly interact with iron-related pathways and will be developed in detail in Parts II and III.

Iron Metabolism

Disruptions in iron metabolism, whether due to deficiency or overload, impair cellular energy production and oxygen handling, leading to the accumulation of reactive oxygen species (ROS). These unstable oxygen-containing molecules damage DNA, proteins, and cell membranes ³. Figure 1.

In iron deficiency, mitochondrial respiration is compromised, causing electron-transport chain leakage and excess ROS generation. Conversely, iron overload promotes ROS formation via the Fenton reaction, in which Fe^{2+} catalyzes hydrogen peroxide into highly reactive hydroxyl radicals ⁴. Both scenarios culminate in oxidative stress, driving tissue injury, reducing therapeutic efficacy, and contributing to the progression of numerous diseases. Figure 2.

Hypoxia-Inducible Factors (HIFs)

Cells depend on hypoxia-inducible factors (HIFs) to gauge oxygen levels and adjust gene expression, which may foster carcinogenesis.

HIF-1 α and HIF-2 α are protein transcription factors that act as cellular "oxygen gauges." Under normal oxygen conditions, and with sufficient iron, HIF- α subunits are hydroxylated by prolyl hydroxylases (iron-dependent

enzymes) and rapidly degraded, keeping intracellular HIF levels very low. When oxygen availability falls or iron-dependent degradation is impaired (for example, due to iron deficiency or oxidative damage), HIF- α escapes degradation, translocates to the nucleus, and dimerizes with HIF- β . This HIF complex then binds to hypoxia response elements in target genes of DNA, turning on pathways that help cells adapt to low-oxygen conditions.

Chronic, abnormally prolonged HIF stabilization disrupts both cellular and extracellular homeostasis by enforcing metabolic programs that favor tumorigenesis. HIF-driven induction of vascular endothelial growth factor (VEGF) promotes angiogenesis, thereby supplying nutrients and oxygen to expanding cellular neoplastic regions. Concurrently, HIF upregulates glycolytic enzymes, shifting metabolism toward glycolysis even in the presence of oxygen, a hallmark of the Warburg effect. Over time, these alterations create a microenvironment conducive to neoplastic proliferation, which can drive the transition from benign, precancerous lesions to malignant tumors.^{5,6,7}

Chronic and or Intermittent Prolonged HIF Stabilization/Activity

Chronic HIF signaling, predominantly mediated by HIF-1 α , drives several premalignant processes unknown to patients that often elude routine clinical detection. First, sustained upregulation of glycolytic enzymes and glucose transporters ensures a continuous supply of ATP and macromolecular precursors (Warburg effect), fostering an environment conducive to neoplastic transformation. Second, persistent HIF-1 α activity may maintain a subpopulation of therapy-resistant cancer stem cells (CSCs) by upregulating stemness-related factors and enabling cellular quiescence ⁸. Finally, HIF-driven expression of immune checkpoint ligands, most notably PD-L1, on both CSCs and differentiated tumor cells blunts cytotoxic T-cell responses, facilitating immune evasion.⁹ Together, these mechanisms link disrupted iron metabolism to chronic inflammation, impaired immune surveillance, and eventual carcinogenesis, underscoring the importance of recognizing both iron deficiency and overload as silent drivers of HIF-mediated tumorigenesis.

High PD-L1 expression, often driven by iron deficiency or hypoxia/HIF activity, is associated with poorer prognosis in lung, breast, renal, gastric, and colorectal cancers.¹⁰ In addition, tumors exhibiting high HIF-1 α expression and intratumoral hypoxia demonstrate significantly reduced overall survival (OS) and progression-free survival (PFS) ¹¹ and show impaired responses to conventional therapies, as hypoxic conditions confer resistance to both radiation and chemotherapy.¹²

In parallel, iron-driven oxidative stress (from too little or too much) activates NF- κ B (nuclear factor kappa-light-chain-enhancer of activated B cells), a master transcription factor in inflammation. NF- κ B induces proinflammatory cytokines, prosurvival signals, and antiapoptotic genes, fostering a chronic inflammatory microenvironment that supports angiogenesis and tumor progression.^{13,14,15}

A third key regulator, Nrf2 (nuclear factor erythroid 2-related factor 2), normally controls antioxidant defenses. Under oxidative stress, Nrf2 translocates to the nucleus to upregulate detoxification and redox-homeostasis genes. However, both iron deficiency (limiting iron-dependent enzyme production) and iron overload (overwhelming Nrf2 with excessive ROS) impair this response.^{16, 17}

The resulting antioxidant failure amplifies damage initiated by HIF and NF- κ B, accelerating DNA injury, genomic instability, and malignant transformation.^{18,19, 20} Figures 3,7.

By linking iron imbalance to these three intersecting pathways —HIF-driven metabolic reprogramming and immune evasion, NF- κ B-mediated inflammation, and compromised Nrf2 antioxidant defenses —clinicians can better understand how subtle iron disturbances promote premalignant changes and identify novel intervention points.

Synergistic Crosstalk of Hypoxia, Inflammation, and Redox Pathways in Tumor Aggressiveness

Tissue hypoxia, micronutrient insufficiency, and inflammation converge to activate HIFs, NF- κ B, and Nrf2, triggering ROS (reactive oxygen species)-mediated oxidative stress, immune evasion, and metabolic reprogramming. This coordinated response fosters a pro-tumor microenvironment characterized by genomic instability, resistance to apoptosis, and diminished responsiveness to conventional therapies, sustaining a pro-tumor inflammatory milieu.^{21,22,23}

For clinicians, understanding how iron metabolism, hypoxia signaling, and chronic inflammation intersect clarifies why certain tumors progress rapidly after initiation and are resistant to treatment (chemotherapy, radiation, and immunotherapy). It may reveal new therapeutic targets within these core pathways. Moreover, iron dysregulation (too much or too little) itself is a key driver of carcinogenesis; its contribution to oxidative stress can compound other latent micronutrient deficiencies, further amplifying pro-tumor oxidative damage.

Relative Reactivity of Major Reactive Oxygen and Nitrogen Species

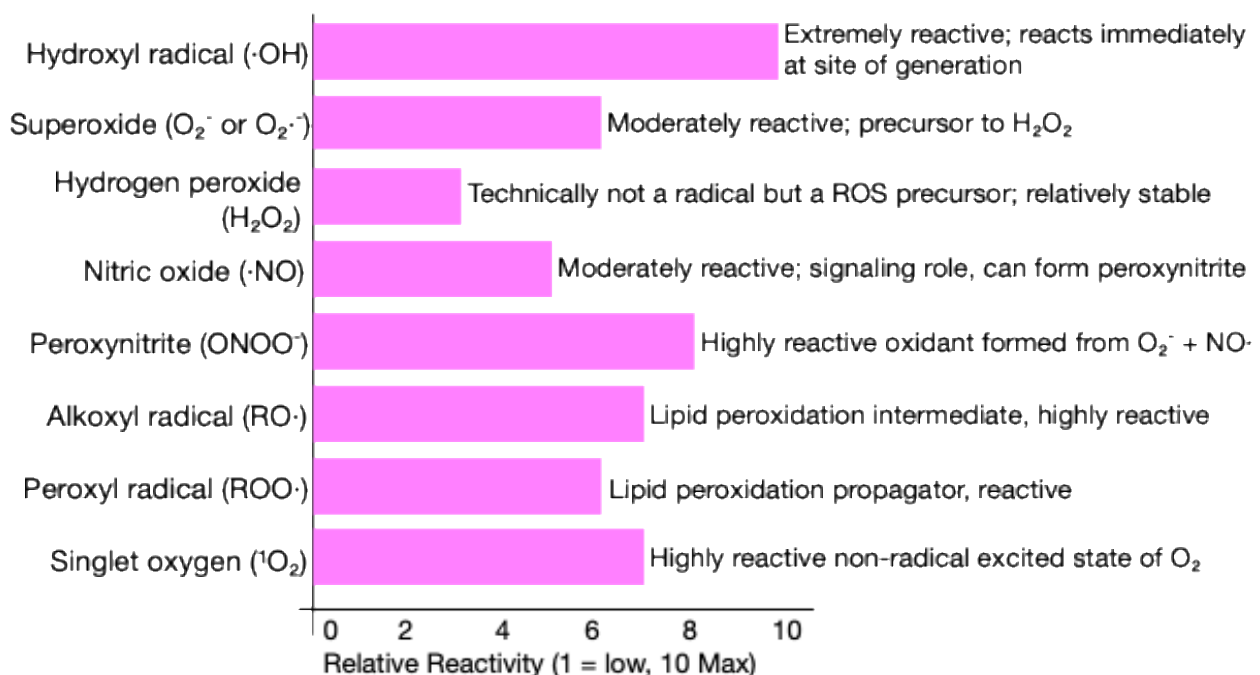


Figure 1: This figure illustrates the relative chemical reactivity of key reactive oxygen species (ROS) and reactive nitrogen species (RNS) implicated in oxidative stress. The hydroxyl radical ($\cdot\text{OH}$) is the most reactive species, causing immediate damage to nucleic acids, lipids, and proteins at the site of formation. While less reactive, Superoxide (O_2^- or $\text{O}_2^{\cdot-}$) serves as a precursor to hydrogen peroxide (H_2O_2), which may contribute to hydroxyl radical formation via Fenton chemistry. Peroxynitrite (ONOO^-), generated from superoxide and nitric oxide ($\cdot\text{NO}$), is also highly reactive and cytotoxic. Peroxyl ($\text{ROO} \cdot$) and alkoxyl ($\text{RO} \cdot$) radicals propagate lipid peroxidation. Although not a radical, singlet oxygen ($^1\text{O}_2$) is an excited state of molecular oxygen with potent oxidative potential. Reactivity rankings reflect relative tendencies to initiate biomolecular damage under physiological conditions. Illustration by Glenn Tisman, M.D.

Iron Deficiency vs. Iron Overload: Divergent and Convergent Mechanisms of Reactive Oxygen Species Generation

Iron Dysregulation and Oxidative Stress Mechanisms: Vitamin C Potentiation of ROS

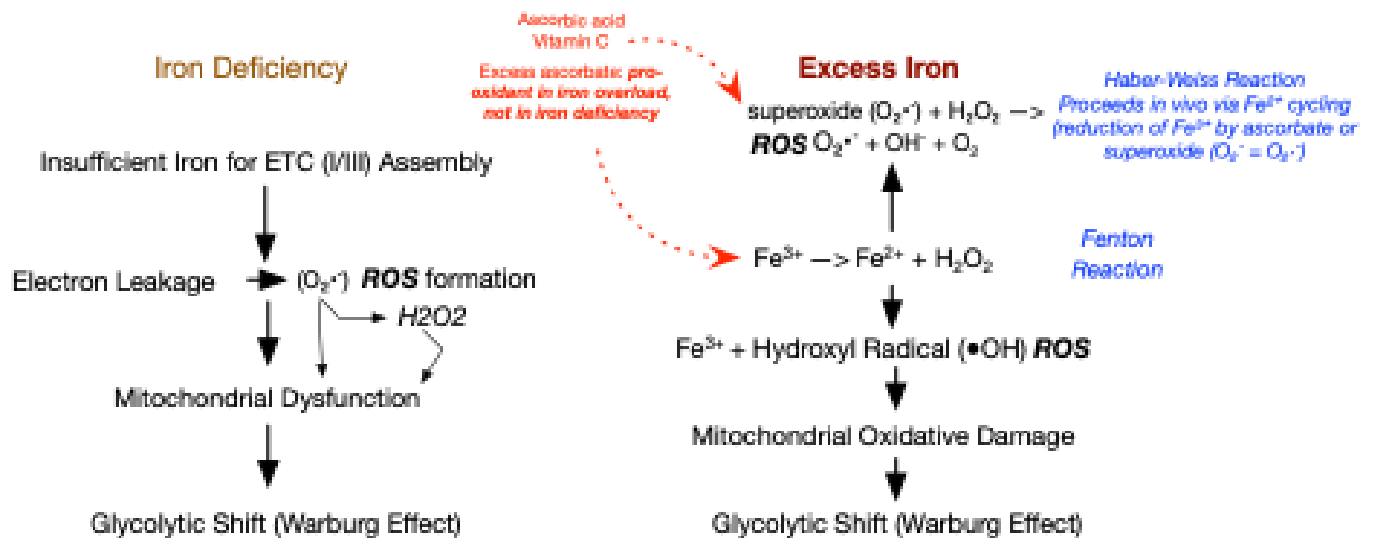


Figure 2 Legend: Iron status shifts cellular redox balance in opposite directions. Iron deficiency disrupts electron-transport-chain assembly, causing electrons to leak and form superoxide and hydrogen peroxide, and forces cells to rely on glycolysis (the Warburg effect). In contrast, iron overload drives Fenton chemistry ($Fe^{2+} + H_2O_2 \rightarrow Fe^{3+} + \cdot OH + OH^-$), amplifying hydroxyl-radical production and triggering broad metabolic reprogramming.

When high doses of vitamin C are present in iron-overloaded cells, ascorbate reduces Fe^{3+} back to Fe^{2+} , fueling a redox cycle that dramatically increases highly destructive $\cdot OH$ formation through the Haber–Weiss reaction (e.g., hereditary hemochromatosis with HFE mutations).^{24, 25} This pro-oxidant amplification does not occur in iron deficiency, where vitamin C remains an effective antioxidant.

Balancing iron levels is therefore crucial, as both deficiency and excess can contribute to oxidative stress. However, iron overload (not deficiency) combined with high ascorbate intake leads to the generation of runaway, highly destructive reactive oxygen species ROS.²⁶ Illustration by Glenn Tisman, M.D.

Clinical Iron Overload/Excess Status Biomarkers: Normal Ranges, Overload Thresholds, Follow-Up Actions

Test	Purpose	Normal Range	Abnormal Threshold & Interpretation	Follow-Up Action
TSAT (Serum Transferrin Saturation)	% of transferrin bound to iron	approx >24<40-45%	≥45%: Indicates iron overload; in men, concern at ≥50%	If ≥45%, obtain HFE genotyping (C282Y/H63D ± S65C)
Serum Ferritin	Reflects total body iron stores	Men: 30–300 ng/mL; Women: (15–200 ng/mL)	>300 ng/mL (men) or >200 ng/mL (women): Suggests iron overload	Combine with TSAT; if both elevated, proceed to genetic testing
Serum Iron	Circulating free iron	Men: 50–150 µg/dL; Women: 35–145 µg/dL	Above upper normal: Supports elevated TSAT interpretation	Repeat fasting; calculate TSAT; assess TIBC if discrepancy
Total Iron-Binding Capacity (TIBC)	Transferrin's iron-binding capacity	250–400 µg/dL	Low TIBC with high TSAT/ferritin: Consistent with hemochromatosis pattern	Interpret alongside TSAT and ferritin

Table 1 Legend: This table presents four routinely measured serum biomarkers used to assess iron status and detect iron overload. For each test, the normal reference range is given alongside the threshold at which iron excess should be suspected. Recommended follow-up actions, such as genetic testing for hereditary hemochromatosis, are specified for abnormal results. The References column provides direct links to peer-reviewed articles that establish or validate these interpretive criteria. Lower limits for normal ranges of serum ferritin vary by the presence of concomitant inflammation and new hepcidin-iron absorption data. Recall that excess ascorbic acid ingestion through the Haber-Weiss and Fenton reactions enhances iron-associated oxidative damage.^{27,28,29,30,31}

Hereditary hemochromatosis is a common genetic disorder of iron metabolism, most frequently associated with mutations in the *HFE* gene. The clinical risk of developing iron overload varies by genotype, with homozygous C282Y mutations carrying the highest

penetrance for iron accumulation and disease expression. Understanding the population prevalence and associated risk of overload for each variant helps guide genetic screening and early intervention strategies (Table 2).

Prevalence of Hemochromatosis Mutations and Risk for Iron Overload

Mutation Type	Prevalence in USA (%)	Iron Overload Risk for Carriers
C282Y Homozygotes	~0.44 (1 in 227 Whites)	High, typical for disease expression
C282Y Heterozygotes	~6.4	Generally low, unless compound heterozygous
H63D Heterozygotes	~16.9	Very low, unless combined with other mutations
S65C Heterozygotes	~1-3	Low, rarely significant
Compound Heterozygotes	Variable, ~1-2%	Moderate to high, especially with excess iron intake

Table 2 Legend: Table 2 summarizes the estimated prevalence of key HFE (High Iron Fe) gene mutations in the U.S. population and their corresponding risk for iron overload. The C282Y mutation is most strongly associated with clinically significant iron overload when homozygous. Heterozygous C282Y or H63D mutations are generally benign unless present in compound heterozygosity, in which case the risk increases, particularly when compounded by dietary or metabolic factors. The S65C variant is less common and rarely associated with clinically relevant iron accumulation. Population data are primarily derived from White non-Hispanic cohorts, where the condition is most prevalent.

Both Subclinical and Clinical Iron excess are underappreciated drivers of malignancy risk and cancer-therapy resistance, even before overt disease manifests.

Although deficiencies in vitamin D, B12, and folate occur in parts of the U.S. population, iron deficiency remains by far the most common micronutrient shortfall.

Iron Deficiency Prevalence

Absolute iron deficiency occurs when total body iron stores are depleted, leading to insufficient iron availability for hemoglobin synthesis. Functional iron deficiency arises when iron stores are adequate but locked away by inflammation-driven hepcidin activity, preventing mobilization of iron for red blood cell production as well as decreased duodenal absorption.

Recent NHANES data report that 14% of U.S. adults have absolute iron deficiency, with another 15% demonstrating functional iron deficiency, together affecting nearly one-third of adults.³² These figures underscore the broad reach of iron deficiency, its frequency of latency (absence of anemia),³³ and its potential impact on multiple clinical outcomes fostering neoplasia (HIF, NF-κB, Nrf2, ROS generation resulting in oxidative stress).

Molecular Consequences of Iron Imbalance

Disruptions in iron metabolism, whether due to deficiency or overload, compromise the redox balance, the equilibrium between pro-oxidant and antioxidant forces that governs cellular oxidative stress. This destabilization,

along with impaired oxygen-sensing pathways, fosters a permissive environment for neoplastic transformation.

Oxidative stress occurs when ROS production surpasses antioxidant defenses, leading to damage in proteins, lipids, and DNA, which promotes mutation and disease progression.³⁴ Hypoxia signaling is activated as HIFs stabilize under oxidative conditions and or iron imbalance, reprogramming metabolism, driving angiogenesis, and impairing immune regulation. These active and intersecting pathways can remain abnormal and clinically silent for years yet progressively remodel the neoplastic or tumor microenvironment of stressed tissues, ultimately fostering insipient dysplasia, which eventually ends in malignant transformation ^{35, 36}

Iron's Dual Role in Carcinogenesis

Paradoxically, both iron deficiency and iron overload contribute to cancer risk: Deficiency limits the activity of iron-dependent antioxidant enzymes with mitochondrial leak of ROS, thereby heightening oxidative damage. Excess catalyzes Fenton chemistry, producing highly destructive reactive oxygen species (ROS), specifically hydroxyl radicals, which further damage cellular macromolecules. Figure 2

Chronic, subclinical iron deficiency and excess often go unnoticed, as much as 28 percent of iron-deficient patients and as many as 75% of hereditary hemochromatosis patients. Inadequate laboratory screening allows pro-oxidative and pro-inflammatory disturbances to continue unrecognized in both iron deficiency and overload. ³⁷

Iron Deficiency Prevalence Across High-Risk Groups

Population	Ferritin Criteria for LID (µg/L)	Other Criteria for LID	Prevalence of LID (%)	Ferritin Criteria for Total Iron Deficiency (µg/L)	Prevalence of Total Iron Deficiency (%)	Prevalence of IDA (%)
Women (18–50 years, nonpregnant)	<30	TSAT <20%; Hb ≥12 g/dL	11.5	<30	16.3	4.8
Adolescent Females (12–21 years)	<25	TSAT <20%; Hb ≥12 g/dL	32.7	<25	38.8	6.1
Men (18–50 years)	<30	TSAT <20%; Hb ≥13 g/dL	7.5	<30	8.7	1.2
Adults (15–49 years, mixed)	<30	TSAT <20%; Hb normal	9–12	<30	14	5
Female Athletes (18–41 years)	≤16	sTfR:log ferritin ≥4.5; Hb normal	29–36	≤16	Not reported	Not reported

Children/Toddlers (1–2 years)	<12	TSAT <20%; Hb normal	6–9	<12	9	3
Pregnant Women	<15	TSAT <20%; Hb ≥11 g/dL	Not well quantified	<15	~12	~12
Older Adults (65+ years)	<30	TSAT <20%; Hb ≥13 g/dL	12–15	<30	15	5–8
Postpartum Women (0–6 months)	<30	TSAT <20%; Hb ≥12 g/dL	~20	<30	25–30	~15
Vegetarian/Vegan Adults	<30	TSAT <20%; Hb normal	15–35	<30	20–35	4–6

Table 3 Legend: This table outlines the estimated prevalence of latent iron deficiency (LID), total iron deficiency (ID), and iron-deficiency anemia (IDA) in U.S. demographic subgroups. LID is identified by low serum ferritin without anemia, while total ID includes LID and anemia cases. Diagnoses are based on ferritin, transferrin saturation, sTfR/log ferritin ratio, and hemoglobin thresholds. Data come from NHANES, cohort studies, athlete screenings, WHO guidelines, and international cohorts with similar conditions. One must be cognizant of the fact that a sole determination of serum ferritin is frequently not alone diagnostic of iron deficiency or overload. For 2024 American Gastroenterological Association Recommendations on diagnosis and therapy of iron deficiency.³⁸ **The cutoff for serum ferritin to diagnose iron deficiency has been raised to less than 45–50 µg/L in the absence of chronic inflammation/cancer.**³⁹

Dangers of Latent Iron Deficiency (LID) and Iron Deficiency with Anemia (IDA)

Even early-stage iron deficiency, before anemia becomes apparent, triggers metabolic oxidative stress, elevates reactive oxygen species (ROS), and stabilizes hypoxia-inducible factors (HIFs). ROS (e.g., $O_2^{\bullet-}$, H_2O_2 , $\bullet OH$) at high levels damage DNA (and epigenetic marks), proteins, and cell membranes, leading to mutations, impaired apoptosis, or necrosis, and are central to aging and carcinogenesis.^{40, 41}

In iron-deficient mitochondria, impaired electron-transport chain activity increases superoxide production, and residual iron participates in Fenton chemistry when H_2O_2 clearance (via catalase and other antioxidants that require normal levels of iron) is compromised, fueling a vicious cycle of mitochondrial injury and ROS accumulation.⁴² Whether in LID or full-blown IDA, these processes reshape cellular energy metabolism, alter transcriptional networks, and impair immune surveillance, laying the groundwork for malignant transformation.^{43, 44}

Synergy with Other Micronutrient Deficiencies and the Theoretical Framework

Deficiencies in B_{12} , folate, and B_6 cause toxic metabolite accumulation (e.g., methylmalonic acid, homocysteine),

amplifying oxidative stress and inflammation. Vitamin D deficiency further compromises immune regulation, facilitating tumor immune evasion. Often clinically silent, these latent deficits still induce DNA and epigenomic damage, favoring an immunosuppressive, therapy-resistant niche of abnormal cells.

Ames's "Triage Theory" posits that subclinical deficiencies reroute scarce resources toward short-term survival at the expense of genomic maintenance, increasing long-term cancer risk even without overt symptoms.⁴⁵ Nutrient compartmentalization, as shown by Das and Herbert⁴⁶ in lymphocyte folate and B_{12} studies, highlights how micronutrient distribution varies by cellular turnover rate (Figure 3). Slowly proliferating cells, such as neurons, epithelial stem cells, or memory immune cells, may experience silent molecular injury due to early deficiency, including impaired DNA repair and redox imbalance. In contrast, rapidly dividing tissues temporarily mask these deficits by preferentially accumulating available nutrients, thereby delaying clinical detection of subclinical micronutrient insufficiency.⁴⁷ Overall, even marginal micronutrient shortages can tip redox signaling toward chronic disease, including cancer.^{48, 49}

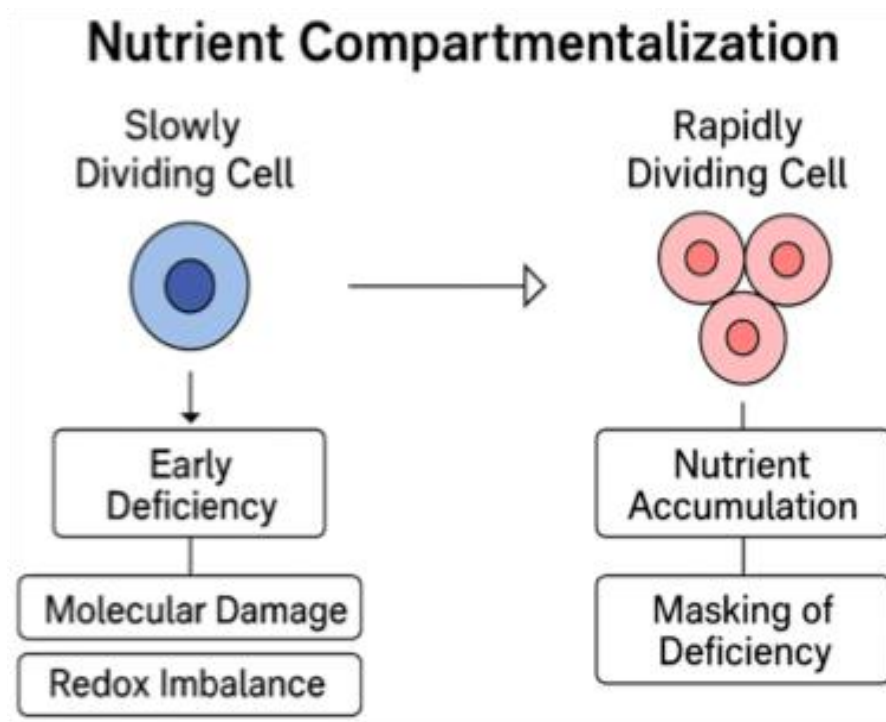


Figure 3 Legend: Nutrient Compartmentalization and Cell-Type Vulnerability in Micronutrient Deficiency
 This schematic illustrates how slowly dividing cells (left), such as neurons and mature lymphocytes, are more vulnerable to chronic intracellular micronutrient deficiency because they have fewer opportunities for nutrient uptake. In contrast, rapidly proliferating cells (right), such as hematopoietic or mucosal cells, frequently re-enter the cell cycle, allowing for more consistent uptake and temporary normalization of intracellular levels. This disparity helps explain why serum markers may appear normal despite persistent subclinical tissue-level deficiency. Illustration by Glenn Tisman, M.D.

Important Consideration Regarding Nutrient Compartmentalization

Figure 3 builds upon foundational work by Das and Herbert,⁴⁶ who demonstrated that lymphocyte folate and B₁₂ levels remain depleted long after serum concentrations normalize, particularly in patients with prior deficiency. The underlying mechanism is related to the cell cycle–linked expression of nutrient transporters, which are primarily active during proliferative phases, such as the S-phase. Slowly dividing or quiescent cells, such as resting epithelial stem cells, neurons, or memory lymphocytes, spend extended periods in non-proliferative states with limited transporter activity, allowing only intermittent and inefficient micronutrient incorporation. When external supply is marginal or fluctuating, these cells accrue chronically low intracellular levels despite transient serum normalization.

In contrast, rapidly cycling cells, which pass through uptake-permissive phases more frequently, can preferentially absorb nutrients and temporarily mask broader systemic deficiencies. This explains the clinical observation that serum folate or B₁₂ may normalize quickly with supplementation, while red blood cell or lymphocyte assays, which reflect long-term intracellular stores, remain low. Persistent symptoms, such as neuropathy, fatigue, or mucosal atrophy, may therefore signal an ongoing tissue-specific deficiency, even in the

presence of seemingly adequate laboratory values. This framework highlights the limitations of serum-only vitamin assessments. It supports the role of intracellular and/or other metabolic biomarkers, such as tHcy and MMA, in detecting functionally relevant micronutrient deficits associated with B₁₂, B₉, and B₆ deficiencies.

Redox Imbalance Leads to Progressive Clinical Symptoms

A redox imbalance, defined as a pathological state in which reactive oxygen species (ROS) production exceeds antioxidant defenses, drives oxidative stress and cellular dysfunction.⁵⁰ This fundamental disturbance underlies both subclinical and overt disease processes, including chronic inflammation, cardiovascular dysfunction, neurodegeneration, and carcinogenesis.⁵¹

In clinical practice, assessing oxidative stress biomarkers, such as malondialdehyde or F₂-isoprostanes, alongside micronutrient status is essential (Table 4). Subclinical deficiencies in iron, vitamin B₁₂, folate, or vitamin D can impair key enzymatic pathways that neutralize ROS and repair oxidative damage.⁵² Identifying these imbalances is particularly important in patients presenting with unexplained fatigue, persistent inflammatory conditions, or early neoplastic, premalignant, or malignant changes, situations in which latent oxidative stress often plays a causative role.

Specimen Types and Clinical Applications of Key Oxidative Stress Biomarkers

Biomarker/Test	Sample Type(s)	Clinical Use
Malondialdehyde (MDA)	Serum/Plasma	Lipid peroxidation marker
F2-Isoprostanes	Plasma or Urine	Oxidative lipid injury (gold standard)
8-Hydroxy-2'-deoxyguanosine (8-OHdG)	Plasma or Urine	Oxidative DNA damage marker
Total Antioxidant Capacity (TAC)	Serum/Plasma	Overall antioxidant status
Glutathione (GSH/GSSG ratio)	Whole Blood	Intracellular redox state
Homocysteine (tHcy)	Plasma (Fasting)	Micronutrient-linked redox marker (B12, B6, folate)
Methylmalonic Acid (MMA)	Serum or Urine	Specific for B12 deficiency
Transferrin Saturation (TSAT)	Serum	Iron availability marker
Soluble Transferrin Receptor (sTfR)	Serum	Absolute Fe def ↑↑ or Functional Fe def = N or slt ↓

Table 4 Legend: This table summarizes the clinical and research available biomarkers used to evaluate oxidative stress and redox imbalance. While no single test is definitive, these markers reflect lipid peroxidation, DNA oxidative injury, antioxidant reserve, and nutrient-linked redox vulnerability.

Elevated levels of total homocysteine (tHcy) and methylmalonic acid (MMA) are valuable early indicators of functionally significant deficiencies in vitamin B12, folate, and vitamin B6 micronutrients that modulate redox homeostasis and mitochondrial function. Markers such as transferrin saturation (TSAT), soluble transferrin receptor (sTfR), and serum ferritin, even in the absence of anemia, also reveal latent iron deficiency that significantly contributes to early oxidative injury.

These tests, as shown in Table 4, often available through specialty or research-focused laboratories, are particularly useful for identifying subclinical redox imbalance in patients with unexplained fatigue, chronic inflammation, or early biochemical signs of carcinogenesis. Targeted assessment of these micronutrient-linked metabolites in clinically asymptomatic individuals may uncover otherwise hidden redox pathologies and guide early preemptive interventions.

Understanding the Chemistry of Iron Excess

Iron Overload: The Fenton and Haber-Weiss Reaction (Figure 2). In states of iron overload, excess ferrous iron (Fe^{2+}) catalyzes the Fenton reaction, in which hydrogen peroxide (H_2O_2) is converted into highly reactive hydroxyl radicals ($\bullet\text{OH}$): $\text{Fe}^{2+} + \text{H}_2\text{O}_2 \rightarrow \text{Fe}^{3+} + \bullet\text{OH} + \text{OH}^-$

Hydroxyl radicals are among the most destructive reactive oxygen species (see Figure 2). They initiate lipid peroxidation, induce DNA strand breaks, disrupt epigenetic marks, and oxidize critical protein thiols. This cascade of oxidative damage drives chronic inflammation, stabilizes hypoxia-inducible factors (HIFs), promotes a metabolic shift that supports Warburg glycolysis in cancer cells, and precipitates genomic instability and cellular senescence, ultimately accelerating malignant transformation.

The Haber-Weiss reaction, as illustrated in Figure 2, further amplifies this process in the presence of iron overload by generating additional hydroxyl radicals from superoxide ($\text{O}_2^{\bullet-}$) and hydrogen peroxide. The superoxide ($\text{O}_2^{\bullet-} = \text{O}_2^{\cdot-}$) originates mainly from mitochondria during cellular respiration but is also produced by specialized enzymes during inflammation.⁵³

Iron overload increases ROS via the Fenton and Haber-Weiss reactions, especially when combined with high-dose vitamin C. Both mitochondrial and inflammatory pathways contribute to the excessive production of superoxide, which fuels the Haber-Weiss and Fenton reactions, ultimately driving ROS-mediated injury ⁵⁴ Figures 1, 2.

Clinically, this is particularly relevant when high levels of ascorbic acid are present, as vitamin C reduces ferric iron (Fe^{3+}) back to ferrous iron (Fe^{2+}), perpetuating the Fenton cycle and enhancing ROS production. This mechanism highlights the potential hazard of excessive vitamin C supplementation in patients with hereditary hemochromatosis, those undergoing repeated blood transfusions, or those with other causes of iron overload.

Key Clinical Insight:

Although classical Fenton chemistry is typically limited in the presence of systemic iron deficiency, mitochondria represent a critical exception. Even when overall iron stores are low, a residual pool of labile iron persists within mitochondria, which are both a source and a target of reactive oxygen species (ROS). Iron deficiency impairs the activity of heme-containing (iron-containing) antioxidant enzymes such as catalase, resulting in H_2O_2 accumulation. Even small amounts of mitochondrial Fe^{2+} can thus fuel Fenton-like reactions, producing dangerous hydroxyl radicals ($\bullet\text{OH}$) and driving localized oxidative injury.

Understanding the Chemistry of Iron Deficiency

Iron imbalance, whether through deficiency or overload, disrupts redox homeostasis and leads to similar downstream consequences, including mitochondrial dysfunction, activation of inflammatory pathways, genomic instability, and promotion of neoplastic transformation. Both extremes ultimately culminate in chronic tissue injury, sustained inflammation, and an elevated risk of malignancy.⁵⁵

In iron deficiency, although the systemic labile iron pool (LIP) is diminished, localized pools of mitochondrial iron may persist. Within mitochondria, this residual iron can participate in Fenton chemistry if hydrogen peroxide (H_2O_2) detoxification is impaired.⁵⁶ Deficits in antioxidant enzymes such as catalase, GPx, and peroxiredoxin

activity, exacerbated by micronutrient shortages (iron, selenium, NADPH precursors) and ongoing oxidative pressure, allow mitochondrial H_2O_2 to accumulate. This fuels Fenton reactions ($\text{Fe}^{2+} + \text{H}_2\text{O}_2 \rightarrow \text{Fe}^{3+} + \cdot\text{OH} + \text{OH}^-$), perpetuating mitochondrial and potentially damaging ROS $\cdot\text{OH}$ generation and redox imbalance.

Figure 2. Because both iron deficiency and iron overload converge on these harmful pathways, impairing antioxidant defenses and enabling iron-driven ROS formation, maintaining balanced iron homeostasis is critical to prevent oxidative tissue damage and reduce cancer risk.

Overt Clinical Tissue Damage Due to Iron Dysregulation

Iron Deficiency

Plummer–Vinson syndrome (PVS) exemplifies clinically silent iron deficiency: early-stage iron depletion may present with minimal or no symptoms, yet, over time, it induces chronic mucosal irritation and dysplasia in the proximal esophagus and/or oropharynx. These microenvironmental alterations, driven by oxidative stress and inflammation, significantly elevate the risk of esophageal or oropharyngeal squamous cell carcinoma.⁵⁷ Although iron replacement therapy reduces the incidence of subsequent malignancies, irreversible tissue damage incurred before treatment may still progress to cancer despite immediate intervention.

Within mitochondria, localized pools of iron can persist even when the systemic labile iron pool (LIP) is diminished. This residual mitochondrial iron participates in Fenton chemistry if hydrogen peroxide (H_2O_2) detoxification is impaired, due, for example, to deficiency of catalase (an iron-requiring enzyme) or other antioxidant enzymes, nutritional imbalances, and heightened redox stress (Figure 2). Mitochondrial injury then leads to increased superoxide ($\text{O}_2^{\cdot-}$) production and ROS leakage. Accumulating H_2O_2 fuels further Fenton reactions, perpetuating mitochondrial damage in a vicious cycle.⁵⁸

Iron Overload

In contrast, iron overload expands the systemic LIP, providing abundant Fe^{2+} substrate for the Fenton reaction. Fe^{2+} reacts with H_2O_2 to generate highly reactive hydroxyl radicals ($\cdot\text{OH}$), markedly increasing oxidative tissue injury and carcinogenic transformation.⁵⁹ Figure 2.

Summary and Clinical Significance

Both iron deficiency and iron overload disrupt redox balance (the dynamic equilibrium between oxidants, such as ROS, and antioxidants that maintains cellular homeostasis and prevents oxidative damage to biomolecules) through distinct yet convergent pathways. Both involve metabolic reprogramming. Warburg glycolysis reprogramming is most efficient in maintaining malignant transformation.

Iron Deficiency Mechanism: Impaired synthesis of iron–sulfur (Fe–S) clusters in mitochondrial complexes I–III causes electrons to leak onto molecular oxygen,

generating excess superoxide ($\text{O}_2^{\cdot-}$). This superoxide undergoes dismutation, either spontaneously or catalyzed by superoxide dismutase, into hydrogen peroxide (H_2O_2), driving oxidative damage despite limited Fenton chemistry under low-iron conditions.⁶⁰ Figure 2.

Iron Overload Mechanism: Excess Fe^{2+} in the expanded LIP catalyzes the Fenton reaction ($\text{Fe}^{2+} + \text{H}_2\text{O}_2 \rightarrow \text{Fe}^{3+} + \cdot\text{OH} + \text{OH}^-$), producing hydroxyl radicals ($\cdot\text{OH}$) that directly attack lipids, proteins, and DNA, thereby exacerbating oxidative stress (Figure 2).

Together, these mechanisms underscore why maintaining iron homeostasis is crucial: physiological ROS levels are required for normal hypoxia-inducible factor (HIF) signaling. At the same time, adequate antioxidant defenses safeguard cellular integrity against oxidative injury.

Warburg Effect and Fenton Chemistry: Metabolic Reprogramming and Redox Shifts

(A Simplified Explanation, see Figures 1 & 2)

The Warburg Effect: Glycolysis Over Oxidative Phosphorylation

Cancer cells frequently favor glycolysis (cytosolic breakdown of glucose) over mitochondrial oxidative phosphorylation (OXPHOS, or oxidative phosphorylation, is the metabolic process in which cells use enzymes to oxidize nutrients, primarily glucose, in the mitochondria, producing ATP as an energy source through the electron transport chain), even when oxygen is plentiful.

However, the hypoxia-inducible factors (HIFs) play a central role in enforcing the Warburg effect by upregulating glycolytic enzymes and glucose transporters, ensuring that cancer cells adapt to hypoxic or nutrient-poor conditions.

The Warburg effect not only enables rapid ATP generation, but it also simultaneously produces biosynthetic precursors necessary for cell growth. A byproduct of this shift is the accumulation of excess lactate, which acidifies the tumor microenvironment, facilitating invasion and immune evasion.⁶¹

Clinical benefit in targeting the Warburg effect is harnessed in the clinic through FDG-PET imaging and metformin therapy, offering tangible diagnostic and therapeutic benefits. FDG-PET imaging detects glycolytically active tumors with a sensitivity of over 90% in breast and lung cancers, guiding staging and treatment adjustments (e.g., altering surgical plans in 36% of NSCLC cases) by identifying occult metastases.⁶² Metformin synergizes with chemotherapy, suppressing lactate production by 40–60% in preclinical models and improving survival outcomes in colorectal and prostate cancers via AMPK-mediated mTOR inhibition.⁶³ These approaches are integrated into clinical protocols, such as using FDG-PET to monitor metformin's metabolic impact in hepatocellular carcinoma trials. A thorough understanding of this shift to glycolysis continues to inform

investigative thinking, leading to the development of therapies and diagnostics.

Warburg Effect Impact on Reactive Oxygen Species (ROS)

Mitochondria are a primary source of ROS during OXPHOS. By downregulating mitochondrial respiration and switching to glycolysis, cancer cells reduce mitochondrial ROS at its source. However, paradoxically, the Warburg shift can still elevate total cellular ROS because glycolytic metabolism disrupts redox balance: levels of NADH and NADPH, essential cofactors for neutralizing ROS, are diminished. This imbalance leads to oxidative stress, resulting in DNA damage and promoting further malignant transformation:^{64, 65} Figure 2.

Iron Deficiency vs. Iron Overload: Divergent Paths to ROS Generation

Iron Deficiency

Impaired iron availability compromises the mitochondrial electron transport chain (ETC), resulting in reduced ATP output and the leakage of electrons to oxygen, which forms superoxide ($O_2^{\bullet-}$). As mitochondria become dysfunctional, cells adapt by ramping up glycolysis to maintain energy production and limit further mitochondrial ROS generation. While this adaptive glycolysis supports short-term survival, prolonged reliance on glycolysis can perpetuate a redox imbalance and lead to cellular injury.⁶⁶

Iron Overload: Excess iron elevates the systemic labile iron pool (LIP), increasing the substrate available for the Fenton reaction: $Fe^{2+} + H_2O_2 \rightarrow Fe^{3+} + \bullet OH + OH^-$

This reaction produces hydroxyl radicals ($\bullet OH$), one of the most reactive and damaging types of ROS. These radicals directly attack DNA, proteins, and lipids, driving oxidative damage and carcinogenic processes.⁶⁷

Normal Cells vs. Malignant Cells: Glycolytic Flexibility

Normal cells

Under stress conditions, such as hypoxia or inflammation, normal cells (e.g., regenerating tissue, immune cells) temporarily increase glycolysis to reduce mitochondrial ROS production. Once the stress resolves, these cells revert to normal mitochondrial OXPHOS, restoring redox balance.

Malignant Cells: In contrast, cancer cells often remain "locked" in a glycolytic state despite having functional mitochondria. Oncogenic mutations (e.g., loss of p53), activation of hypoxia-inducible factors (HIFs), and overexpression of transcription factors like MYC enforce a persistent glycolytic program.

The permanent shift toward glycolysis offers three key advantages

Immediate ROS Reduction: By relying on glycolysis rather than OXPHOS, cells limit mitochondrial electron leakage and acute ROS spikes, thereby aiding survival under stress.

Chronic Redox Imbalance: Lower NADH/NADPH levels weaken antioxidant defenses, allowing ROS to build up over time and drive DNA damage and genomic instability.

Biosynthetic Support & Apoptosis Resistance:

Glycolytic intermediates feed nucleotide, amino acid, and lipid synthesis for rapid proliferation, while reduced mitochondrial signaling limits apoptosis.⁶⁸

Integrated Consequences for Carcinogenesis

When combined, the Warburg effect and iron-driven Fenton chemistry produce a redox environment that favors DNA damage, oncogenic signaling, and genomic instability. In iron-deficient contexts, mitochondrial ETC defects and compensatory glycolysis drive a vicious cycle of ROS leakage and Fenton reactions. In iron-overloaded settings, abundant Fe^{2+} accelerates hydroxyl radical production. Both scenarios disrupt the equilibrium of oxidative stress, overwhelming antioxidant defenses and facilitating malignant transformation. Maintaining iron homeostasis and balanced energy metabolism is, therefore, critical to preserving redox balance and preventing carcinogenesis.

Iron Metabolism and Hypoxia-Inducible Transcription Factors (HIFs)

Transcription factors are master regulators of gene expression that coordinate cellular responses to environmental stressors, including hypoxia, nutrient deprivation (such as iron, folate, B12, B6, homocysteine, methylmalonic acid, and vitamin D), and oxidative imbalance. Among these, the hypoxia-inducible factors (HIFs), particularly HIF-1 α and HIF-2 α , serve as central mediators of metabolic and angiogenic adaptation. Their activity is tightly regulated at the post-translational level through iron- and oxygen-dependent hydroxylation by prolyl hydroxylase domain (PHD) enzymes, which are responsible for limiting the activity of HIFs. Figure 8.

In iron-replete, normoxic conditions, PHDs hydroxylate HIF- α subunits, targeting them for relatively rapid degradation via the von Hippel–Lindau (VHL) E3 ubiquitin ligase complex Figures 4. However, iron deficiency impairs the PHD hydroxylation process, even when oxygen is sufficient, allowing HIF- α 1 or HIF- α 2 transcription proteins to escape degradation and accumulate within the nucleus. Figures 4, and 8. This state of "pseudohypoxia" results in the prolonged transcriptional activation of numerous genes involved in glycolysis, angiogenesis (e.g., VEGF), immune modulation, and cell survival. Sustained stabilization of HIF-2 α has been associated with the development of a stem-like cellular phenotype Figure 6, characterized by therapy resistance (both chemotherapeutic and radiation) and increased cell surface PD-L1, which enhances immune resistance and tumor progression. This highlights the critical role of iron as a metabolic gatekeeper in transcriptional reprogramming and the progression of oncogenic transformation.

Stabilization of Hypoxia-Inducible Factors (HIFs) in Iron Deficiency: Iron is a key cofactor for prolyl hydroxylase domain (PHD) enzymes, which hydroxylate

HIF- α subunits (HIF-1 α /2 α) under normal oxygen and iron conditions, targeting them for VHL-mediated ubiquitination and degradation (Figures 5 and 8). In iron deficiency, insufficient Fe²⁺ impairs PHD activity, reducing HIF- α hydroxylation and allowing HIF- α to accumulate even in the presence of normal oxygen, triggering "pseudohypoxic" signaling.

Persistent HIF activation shifts metabolism toward glycolysis (the Warburg effect), drives VEGF-mediated angiogenesis, inhibits apoptosis, and facilitates immune evasion, collectively fostering tumor survival and progression. Switching between HIF-1 α and HIF-2 α expression further supports the properties of cancer stem cells, conferring resistance to radiotherapy, targeted chemotherapy, and immunotherapy.

The HIF-1 α and HIF-2 α Metabolism Switch During Normoxia and Normal Iron Supply vs. Intermittent to Prolonged Anoxia and Iron Insufficiency/Deficiency

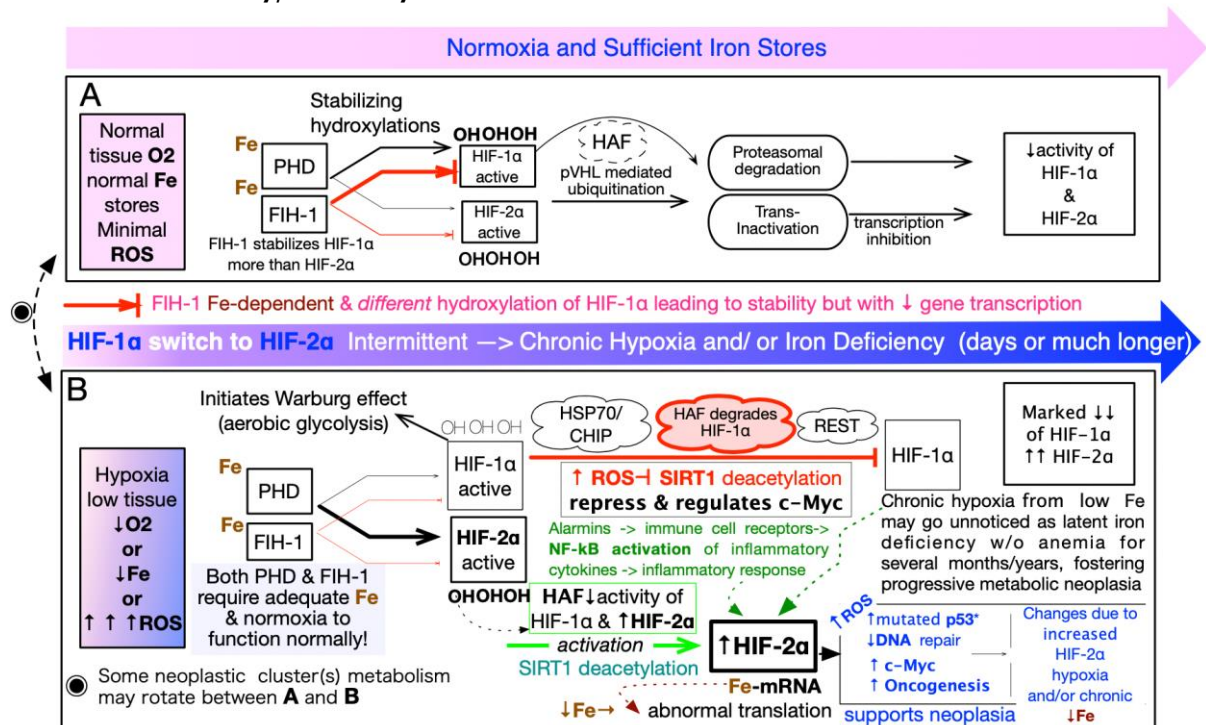


Figure 4 Legend: Panel A: Normoxia and Sufficient Iron Stores and Panel B: Hypoxia, Reduced Tissue O₂, or Iron Deficiency. Illustration by Glenn Tisman, M.D.

As revealed in Figure 4 Panel A, **under normal oxygen tension and adequate iron availability**, Fe-dependent prolyl hydroxylases (PHDs) and Factor Inhibiting HIF-1 (FIH-1) enzymatically hydroxylate HIF-1 α . PHD-catalyzed prolyl hydroxylation and FIH-1-mediated asparagine hydroxylation both require Fe²⁺ as a cofactor. Hydroxylated HIF-1 α is recognized by HAF (Hypoxia-Associated Factor) and targeted for ubiquitination and proteasomal degradation via the von Hippel-Lindau protein (pVHL) pathway, thus suppressing its transcriptional activity. **In this setting, HIF-1 α remains destabilized, HIF-2 α activity is limited, and hypoxia-responsive gene expression is repressed, maintaining cellular homeostasis and restricting neoplastic transformation.**

Panel B of Figure 4 reveals the effects of Hypoxia, Reduced Tissue O₂, or Iron Deficiency on HIF metabolism.

When tissue oxygen or iron levels fall (due to hypoxia or iron deficiency), the activity of PHDs and FIH-1 is compromised, leading to diminished HIF-1 α hydroxylation and subsequent stabilization of HIF-1 α . Stabilized HIF-1 α translocates to the nucleus, activating genes involved in angiogenesis, glycolysis (Warburg effect), and inflammatory responses (via NF- κ B and

cytokine signaling). Elevated ROS and SIRT1 deacetylation further potentiate HIF-1 α activity, while HAF can selectively degrade HIF-2 α or, under chronic stress, facilitate HIF-2 α predominance. **Iron deficiency also impairs Fe-mRNA translation, undermining DNA repair capacity and favoring genomic instability, including the accumulation of dysfunctional p53.**

Tumorigenic Consequences and the HIF-1 α /HIF-2 α Stabilization Figure 4 Panel B

As tumors progress, intermittent hypoxia and iron deficiency give way to chronic HIF-driven adaptations: initially, HIF-1 α promotes Warburg glycolysis and short-term survival, but over time, HIF-2 α takes over, enforcing long-term metabolic reprogramming, stemness, and immune evasion. These changes, along with uneven oxygen, iron, and extracellular matrix (ECM) distribution, establish distinct niches, ranging from well-oxygenated regions to hypoxic or anoxic zones with necrosis, supporting the survival of premalignant or malignant cells.

A more general rendering of the carcinogenic metabolic milieu is illustrated in Figure 8, illustrating the integrated model of neoplastic transformation.

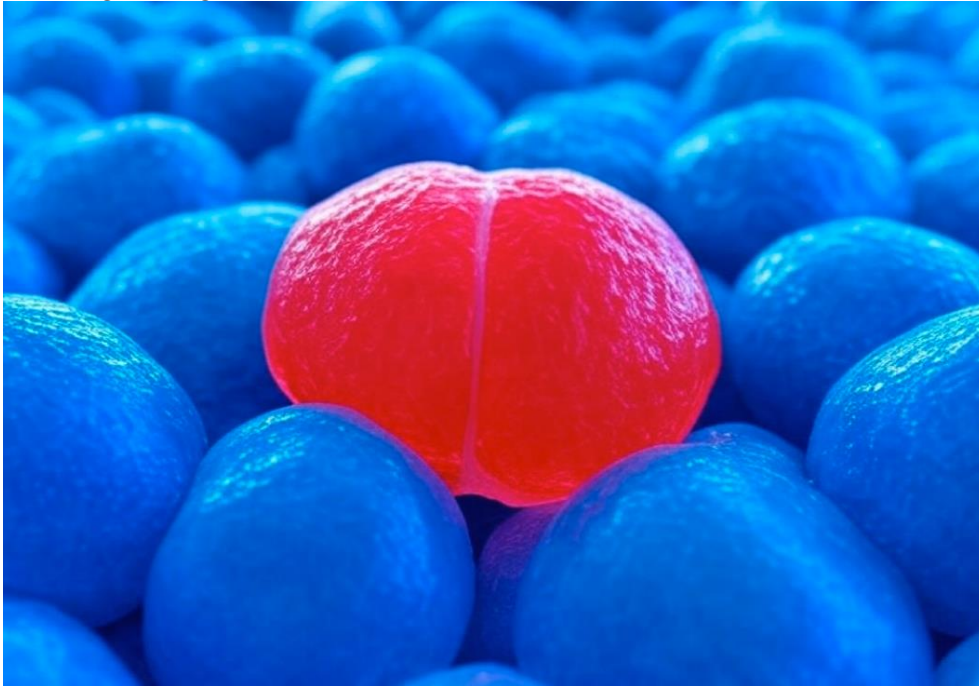
Cancer Stem Cell Dividing Among Committed Cells

Figure 5 Legend: A single dividing cancer stem-like cell (pink/red) is depicted among a field of non-dividing, differentiated, committed tumor cells (blue). This illustration symbolizes the unique capacity of HIF-supported cancer stem cells to self-renew and repopulate the tumor mass, conferring resilience and resistance to both chemotherapy and radiation therapy, and driving disease recurrence and metastasis. Illustration by Glenn Tisman, M.D.

Cancer Stem Cells

Hypoxia-inducible factors (HIF-1 α /2 α) stabilize under low oxygen, migrate to the nucleus, and activate genes that maintain a stem-cell-like state. HIF-2 α upregulates Oct4 and Nanog to preserve a self-renewing CSC subset, while HIF-1 α shifts metabolism toward glycolysis, supplying biosynthetic precursors and reducing oxidative stress, ensuring CSC survival in nutrient-poor, hypoxic regions.^{69,70} Figure 5, HIFs also promote epithelial–mesenchymal transition (EMT), granting CSCs mobility, apoptosis resistance, and immune evasion.⁷¹ Additionally, HIF-driven signals recruit endothelial cells and fibroblasts to form protective niches via Notch–Jagged interactions, further shielding CSCs from immune clearance. Through

these coordinated mechanisms, stemness gene induction, metabolic reprogramming, EMT, and niche formation, CSCs sustain tumor growth, resist therapy, and drive metastasis. Figure 5

Tumor niches are highly dynamic, adapting to changes in size, oxygen gradients, nutrient levels, and therapies (Figure 6). This adaptability supports CSC survival under metabolic stress and can reprogram non-stem tumor cells into stem-like states, driving plasticity and therapy resistance. Although niches may appear histologically as cell clusters, their key feature is functional: they maintain stemness, promote malignant progression, and shield cells from treatment. Targeting these niche regulators could disrupt tumor hierarchies and improve outcomes.

Tumor Stem Cell Niche Microenvironment

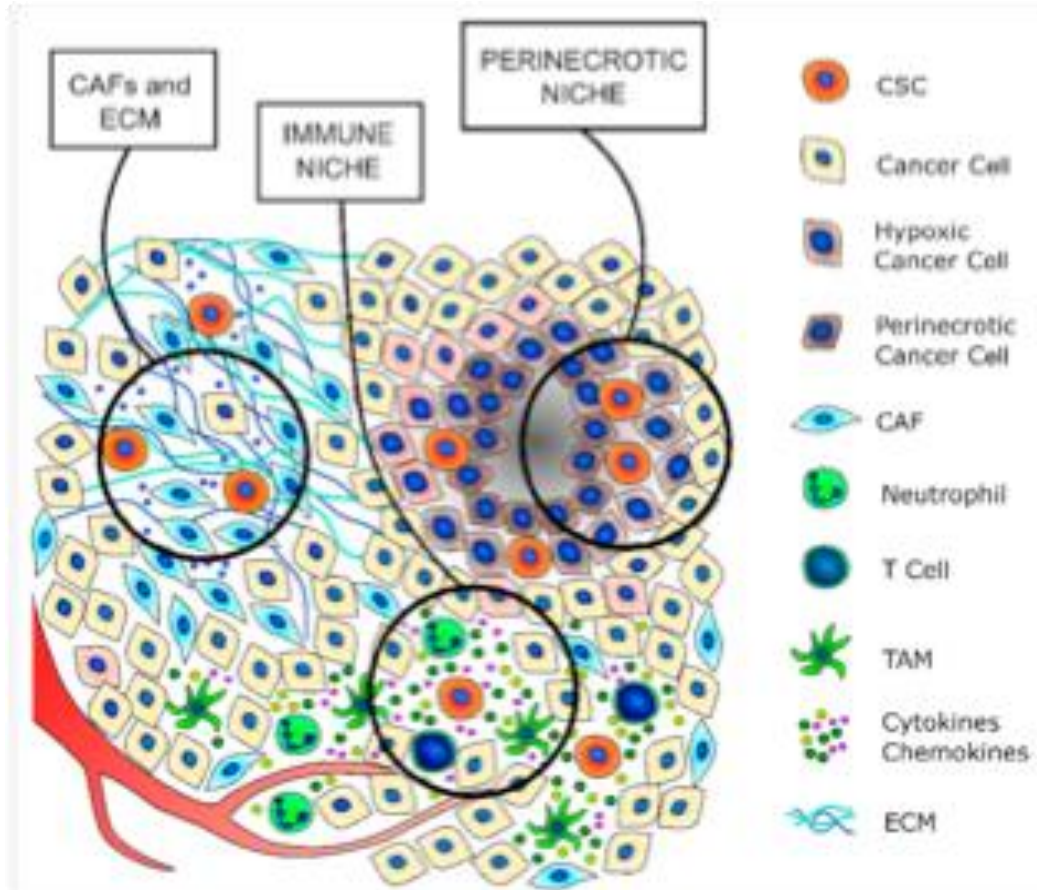


Figure 6 Legend: Cancer stem-like cells (CSCs) reside in specialized tumor microenvironments, known as niches, where multiple factors support their commitment, as well as their maintenance and self-renewal. Key components include: Cancer-associated fibroblasts (CAFs), which remodel the extracellular matrix (ECM) and secrete factors that induce or reinforce stem-like phenotypes in cancer cells; Immune cells (e.g., T cells, tumor-associated macrophages, neutrophils), which regulate CSC behavior through cytokine and chemokine signaling; and Hypoxic regions, where activation of hypoxia-inducible factors (HIFs) sustains and amplifies CSC properties by both maintaining existing CSCs and promoting dedifferentiation of non-stem cancer cells into a stem-like state. Adapted from Grassi ES et al. *J Clin Med.* 2021;10(7):1455.⁷²

Hypoxia-inducible factors (HIF-1 α and HIF-2 α) regulate cancer responses to low oxygen and stress, with activation varying by tumor type, stage, and microenvironment. HIF-1 α dominates in acute hypoxia, driving glycolysis (e.g., via PDK1) and inflammation (IL-6, TNF- α) to boost ATP, proliferation, and DNA-damaging ROS.⁷⁷ HIF-2 α prevails in chronic hypoxia or iron deficiency, collaborating with c-Myc to enhance stemness, therapy resistance, and immune evasion.⁷³ c-Myc, an oncogene, amplifies cell cycle progression and nucleotide synthesis, fueling rapid tumor growth.

Activation isn't universal: HIF-1 α leads in early tumors (e.g., colon carcinoma, rapidly growing triple negative breast cancers), while HIF-2 α drives advanced stages (e.g., VHL-deficient RCC). Hypoxia duration, tumor genetics (e.g., VHL mutations), and stressors (e.g., oxidative stress) dictate dominance. HIF-1 α targets metabolic genes. These genes collectively enable cancer cells to shift from oxidative phosphorylation to glycolysis (the Warburg effect), increase glucose uptake, and store energy in forms such as glycogen and lipids, thereby ensuring survival and proliferation in hypoxic environments. Tumor genetics, such as VHL mutations, can further stabilize HIF-1 α , thereby amplifying the expression of these metabolic genes. Meanwhile, stressors like oxidative stress may modulate their activity

through additional pathways. HIF-2 α focuses on stemness and angiogenesis. These context-specific roles shape tumor progression and therapy response.^{74,75}

Immunostaining Pathology Biopsies for Hypoxia-Inducible Factors: Progression from Normal Neoplasia to Malignancy

Immunohistochemical studies reveal that hypoxia-inducible factors (HIF-1 α and HIF-2 α) are expressed not only in malignant tumors but also in premalignant lesions, signaling early activation of hypoxia-related pathways in cancer development.⁷⁶ Expression increases progressively from normal tissue to dysplasia to carcinoma in organs like the pancreas, cervix, prostate, stomach, and oral cavity, driving metabolic shifts and angiogenesis that promote tumor initiation.^{81,82} As tumors advance, a shift often occurs toward HIF-2 α dominance, which supports cancer stem cell maintenance, enhances angiogenesis, and modulates immune responses, contributing to tumor aggressiveness and treatment resistance.^{77,78} HIF-1 α primarily responds to acute hypoxia, while HIF-2 α sustains chronic hypoxic adaptation, highlighting their distinct roles in the battle between normalcy and malignant transformation. In DCIS of the breast, research supports the role of HIF-1 α in the progression of DCIS to invasive breast cancer, particularly in HER2-negative cells, driven by

inflammatory stimuli via the NF- κ B/COX2 pathway.⁷⁹ Clinical observations of HIF-1 α activity in "hot-spots" further highlight its potential as a progression marker. "Hotspots" in the context of DCIS and HIF-1 α refer to specific areas within the tumor tissue where there is a high level of HIF-1 α activity. These are regions, often near dense stromal infiltration, where basal-like cells show intense expression of HIF-1 α , as seen in immunohistochemical studies.

In papillary thyroid carcinoma and pancreatic ductal adenocarcinoma, co-expression of HIF-1 α and HIF-2 α correlates with aggressive features like lymph node metastasis and higher tumor grade, marking poor prognosis.^{80, 81} These factors serve as biomarkers and therapeutic targets.

Inhibiting HIF pathways disrupts tumor hypoxia adaptation, boosting chemotherapy efficacy, while enhancing ROS exploits tumor cells' sensitivity, inducing apoptosis.⁸²

Arsenic trioxide (ATO) raises ROS in acute promyelocytic leukemia, achieving >80% clinical remission.⁸³ Doxorubicin (Adriamycin), an anthracycline, generates

ROS to kill breast cancer and lymphoma cells, improving response rates and survival.⁸⁴

Standard Histopathology Staining of Tumor Specimens Frequently Mirrors Past Metabolic Stress

Histopathologic evidence of tumor necrosis in biopsy or resection specimens is widely recognized as a marker of tumor aggressiveness and adverse prognosis. Necrosis often arises when rapid cellular proliferation exceeds the tumor's vascular capacity, resulting in hypoxia, metabolic stress, and eventual cell death, particularly in poorly perfused tumor cores. This process is frequently observed in high-grade malignancies, such as glioblastoma,⁸⁵ clear cell renal carcinoma,^{86, 87} non-small cell lung carcinoma,⁸⁸ and triple-negative breast cancers.⁸⁹ Tumor necrosis also reflects the consequences of disorganized or insufficient neoangiogenesis, a hallmark of aggressive tumors characterized by hypoxic signaling and aberrant vascular architecture.⁹⁰ Moreover, necrotic regions release pro-inflammatory and pro-angiogenic mediators that reshape the tumor microenvironment, potentially facilitating immune evasion and further malignant progression.⁹¹ Several tumor grading systems, including the Fuhrman and Nottingham classifications,⁹² incorporate the extent of necrosis as a histological determinant of grade, linking it directly to clinical behavior, metastatic potential, and patient survival outcomes.⁹³

Iron Imbalance Amplifies Oxidative Stress and May Synergize with Other Micronutrient Deficiencies to Multiply the Oxidative Stress Drive to Neoplasia and Carcinogenesis

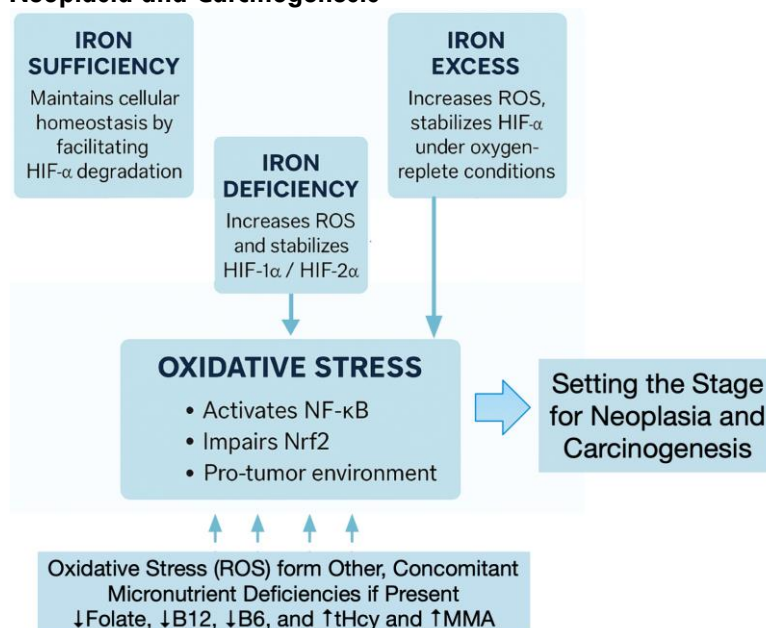


Figure 7 Legend: Effects of Iron Imbalance on Oxidative Stress and Tumor Biology. This diagram illustrates how both iron deficiency and iron excess lead to the accumulation of reactive oxygen species (ROS) and sustained oxidative stress, albeit through distinct mechanisms: primarily mitochondrial dysfunction in deficiency and Fenton chemistry in overload. Both conditions stabilize HIF-1 α and HIF-2 α , activate NF- κ B-mediated inflammatory signaling, and suppress the Nrf2 antioxidant defense system, creating a microenvironment that favors tumor initiation and progression. Illustration by Glenn Tisman, M.D.

These processes are often silent and subclinical, developing over long periods in patients with no overt symptoms and normal routine laboratory findings. The risk is further magnified when other micronutrient deficiencies Figures 7, 8 and Table 5, including folate, vitamin B12, and B6, vitamin D, or elevations in

methylmalonic acid (MMA) and homocysteine (tHcy), are also present, each of which can independently contribute to oxidative, mitochondrial, and epigenetic stress.⁹⁹⁻¹⁰¹ Together, these overlapping biochemical abnormalities converge to generate a permissive environment for neoplastic transformation and carcinogenesis.¹⁰²⁻¹⁰⁶

Potential Points of Synergism and/or Interference of Cellular Oxidative Stress Related to Common Micronutrient Deficiencies and Their Resulting Toxic Metabolites (Homocysteine and Methylmalonic acid)

Feature / Pathway	Vitamin B12 Deficiency	Folate Deficiency	Vitamin B6 Deficiency	Iron Deficiency	Iron Overload	Vitamin D Deficiency
Primary toxic metabolites	↑ MMA, ↑ Homocysteine (tHcy)	↑ tHcy	↑ tHcy, ↑ ROS (indirect)	No direct toxic metabolites	↑ Hydroxyl radicals (Fenton reaction)	↑ Inflammatory ROS via NF-κB activation
Key enzymes affected	↓ Methionine synthase, Methylmalonyl-CoA mutase	↓ Methionine synthase (↓ 5-methyl-THF)	↓ CBS, ↓ SHMT	↓ Ferritin, ↓ Transferrin saturation	↓ ETC complexes I-IV, ↑ HO-1	↓ 1-α-hydroxylase, ↓ VDR-mediated transcription
Glutathione (GSH) effect	↓ GSH due to ROS overproduction	↓ GSH due to ROS overproduction	↓ GSH due to impaired transsulfuration	↓ GSH due to increased ROS	↓ GSH via persistent ROS and Fe-catalyzed ferroptosis	Mild GSH depletion (indirect via inflammation)
NADPH consumption	↑ due to ROS detoxification demands	↑ via antioxidant enzyme burden	↑ NADPH use, regeneration	↑ NADPH consumption due to ROS	↑ NADPH demand from ongoing oxidative stress	↑ ROS burden without strong NADPH adaptation
Methylation capacity (SAM/SAH)	↓ SAM, ↑ SAH → Global hypomethylation	↓ SAM, ↑ tHcy → impaired methylation and use	↓ PLP affects SAM synthesis and use	No direct effect	No direct effect (ROS may impair enzymes)	No direct effect (inflammation may impair)
Antioxidant gene expression	↓ transcription via impaired methylation	↓ from unstable promoters	↓ enzyme activation for GR/SOD/CAT	↑ Ferritin, ↑ TfR; ↑ SOD in response to ROS	↑ HO-1, ferritin; SOD may be overwhelmed	↓ regulation of GR/SOD via VDR
Mitochondrial ROS leakage	↑ (via MMA inhibition of ETC Complex II)	↑ via ETC stress from tHcy toxicity	↑ via impaired ROS scavenging in mitochondria	↑ due to impaired ETC function (Fe-S cluster deficiency) ↓ CAT activity	↑ Fenton reaction from constant redox cycling of Fe ²⁺ /Fe ³⁺	↑ via cytokine-mediated mitochondrial stress
Lipid peroxidation	↑ from tHcy/MMA-induced ROS	↑ from tHcy auto-oxidation	↑ due to inadequate ROS neutralization	↑ due to increased ROS	↑ from lipid oxidation (malondialdehyde, 4-HNE)	↑ from IL-6/Th17 axis
Unique consequence	Neurological damage (e.g., neuropathy)-subacute combined degeneration	Uracil misincorporation, strand breaks	Loss of >140 PLP-requiring redox enzymes	Anemia, fatigue, impaired oxygen transport	Fibrosis, ferroptosis, organ damage	Impaired immune clearance, ↑ HIF-1α, ↑ angiogenesis

Table 5 Legend: Redox-Centric Pathophysiologic Consequences of Select Micronutrient Imbalances

This comparative table outlines how deficiencies in vitamin B12, folate, vitamin B6, vitamin D, and iron, as well as iron overload, disrupt cellular redox homeostasis through distinct yet overlapping biochemical mechanisms that may be synergistic. Each row represents a specific metabolic or genomic vulnerability implicated in the early stages of carcinogenic transformation.¹⁰²⁻¹⁰⁵

Integrated Mechanistic Model of Neoplastic Transformation

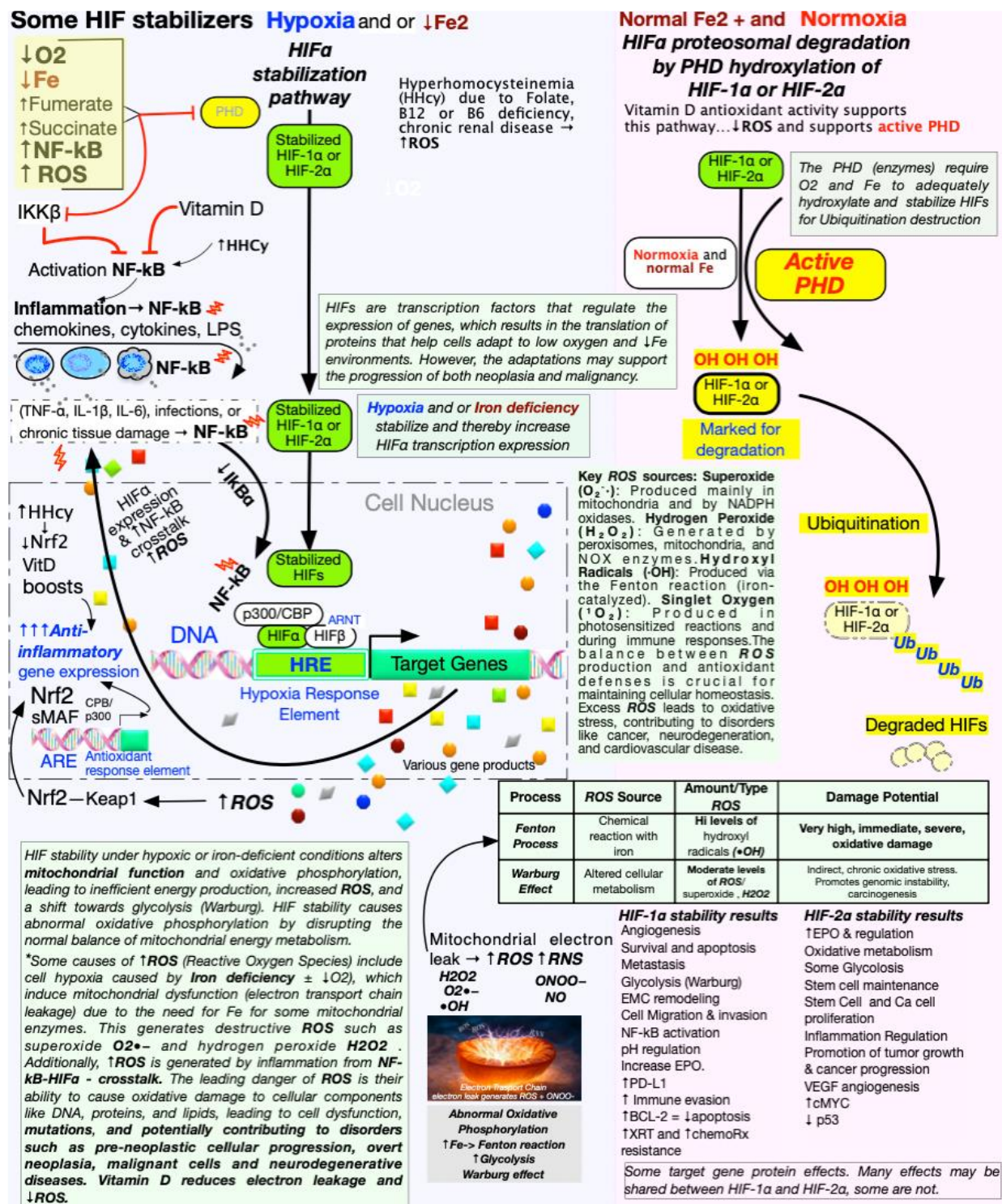


Figure 8 Legend: This figure shows a systems-level model in which chronic, subclinical deficiencies of key micronutrients (iron, folate, vitamin B12, vitamin B6, vitamin D) create an occult pro-neoplastic milieu by sustaining oxidative stress, inflammatory signaling, and hypoxia-inducible factor (HIF) stabilization over years. These ROS-driven carcinogenic pathways remain clinically silent because routine chemistries and blood counts are often normal or only mildly abnormal; even small shifts within the reference range toward abnormal should prompt further evaluation and early intervention. Illustration by Glenn Tisman, M.D.

Figure 8 illustrates stimuli, including low oxygen, iron deficiency, succinate accumulation, and increased ROS, that stabilize HIF-1α and HIF-2α by inhibiting PHD-dependent hydroxylation. Folate and B12 deficiencies indirectly raise homocysteine levels, which in turn increase ROS and inflammatory signaling via NF-κB activation. Vitamin D is shown to modulate both inflammation and mitochondrial ROS. The upper right region illustrates normal iron and oxygen conditions, HIFs undergo

hydroxylation by active PHD enzymes, followed by ubiquitination and proteasomal degradation. When PHD activity is impaired, HIFs accumulate and escape degradation. The center panel reveals stabilized HIFs binding ARNT and co-factors (e.g., p300/CBP), leading to transcription of genes related to angiogenesis, glucose metabolism, immune modulation, and stem cell maintenance. Crosstalk with NF-κB and ROS amplifies this transcriptional program. Nrf2, often activated in parallel

by ROS, modulates antioxidant response elements. **The bottom-left** area illustrates how mitochondrial dysfunction caused by iron or oxygen limitation increases ROS and RNS generation, thereby promoting further HIF stabilization and inflammatory cellular and DNA damage. This contributes to a self-sustaining cycle of oxidative stress and transcriptional dysregulation. **The bottom right area** reveals HIF-1 α and HIF-2 α target gene programs, highlighting their overlapping and divergent roles in cell survival, metabolism, erythropoiesis, and resistance to apoptosis. **The lower table** summarizes ROS contributions to various biochemical pathways and their associated cellular outcomes. This integrated model reveals how even modest disruptions in micronutrient status may initiate chronic inflammation, redox imbalance, and adaptive transcriptional programs that collectively promote benign neoplasia, which may progress to malignant transformation and resistance to both chemo- and radiation therapy.

Conclusion

This first installment, Part I, sounds an urgent alarm for clinicians to investigate latent iron deficiency and coexisting subclinical micronutrient deficiencies (e.g., folate, vitamin B12, vitamin B6, vitamin D) as hidden

drivers of oxidative stress, mitochondrial dysfunction, and early neoplasia, and as potential tumor initiators. Iron's dual role, in which both deficiency and excess disrupt redox balance, demands a paradigm shift in oncology nutritional assessment. These abnormalities synergize to activate HIF and NF- κ B while weakening antioxidant defenses, silently remodeling the tumor microenvironment **long before symptoms appear**. Clinicians must therefore adopt comprehensive micronutrient screening and act on reproducible trends, even within the reference interval, to detect and address these risks early and potentially interrupt premalignant changes or irreversible epigenetic lock-in that can lead to carcinogenesis.

Part II will expand this mechanistic foundation to encompass the complex metabolic and epigenetic processes through which redox imbalance and HIF-driven pathophysiology manifest in patients, using case-based and real-world examples to show how these patterns develop years before a formal cancer diagnosis. Part III will then offer a practical guide for identifying how subtle changes in routine laboratory biomarkers, often reported as "within normal limits," can reveal early subclinical micronutrient imbalances that promote pro-neoplastic biology and therapeutic resistance.

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Abbreviations Used

Abbr.	Definition	Abbr.	Definition	Abbr.	Definition
8-OHdG	8-hydroxy-2'-deoxyguanosine	HIF-1α	Hypoxia-inducible factor-1 alpha	PFS	Progression-free survival
AMPK	AMP-activated protein kinase	HIF-2α	Hypoxia-inducible factor-2 alpha	PHD	Prolyl hydroxylase domain (enzyme)
ARNT	Aryl hydrocarbon receptor nuclear translocator	HO-1	Heme oxygenase-1	PLP	Pyridoxal 5'-phosphate
ATP	Adenosine triphosphate	ID	Iron deficiency	PVS	Plummer–Vinson syndrome
CAT	Catalase	IDA	Iron-deficiency anemia	RNS	Reactive nitrogen species
CBC	Complete blood count	IL-6	Interleukin-6	ROS	Reactive oxygen species
CBS	Cystathionine β-synthase	LID	Latent iron deficiency	SAM	S-adenosylmethionine
CBP	CREB-binding protein	LIP	Labile iron pool	SAH	S-adenosylhomocysteine
CSC(s)	Cancer stem cell(s)	MDA	Malondialdehyde	sTfR	Soluble transferrin receptor
DNA	Deoxyribonucleic acid	MMA	Methylmalonic acid	SHMT	Serine hydroxymethyltransferase
ETC	Electron transport chain	mTOR	Mechanistic target of rapamycin	SOD	Superoxide dismutase
FDG-PET	18F-fluorodeoxyglucose positron emission tomography	NADH	Nicotinamide adenine dinucleotide (reduced)	TAC	Total antioxidant capacity
Fe–S	Iron–sulfur (cluster)	NADPH	Nicotinamide adenine dinucleotide phosphate (reduced)	tHcy	Total homocysteine

Abbr.	Definition	Abbr.	Definition	Abbr.	Definition
GPx	Glutathione peroxidase	NF-κB	Nuclear factor κB	TfR	Transferrin receptor
GR	Glutathione reductase	NHANES	National Health and Nutrition Examination Survey	Th17	T helper 17 cell
GSH	Reduced glutathione	NSCLC	Non-small cell lung cancer	TIBC	Total iron-binding capacity
GSSG	Oxidized glutathione (glutathione disulfide)	Nrf2	Nuclear factor erythroid 2-related factor 2	TSAT	Transferrin saturation
Hb	Hemoglobin	OS	Overall survival	VEGF	Vascular endothelial growth factor
H₂O₂	Hydrogen peroxide	OXPPOS	Oxidative phosphorylation	VDR	Vitamin D receptor
HFE	High-iron gene associated with hereditary hemochromatosis	PD-L1	Programmed death-ligand 1	WHO	World Health Organization
HIF(s)	Hypoxia-inducible factor(s)				